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7 This question is about organic compounds.

(a) Butane reacts with chlorine in a photochemical reaction.



(i) State the meaning of the term photochemical.

caused by violet light (1)

(ii) An organic compound with the formula $\text{C}_4\text{H}_9\text{Cl}$ is formed when one molecule of butane reacts with one molecule of chlorine.

Draw the displayed formula of two possible structural isomers with the formula $\text{C}_4\text{H}_9\text{Cl}$ formed in this reaction.

1. chlorobutane

2. 2-chlorobutane

✓



(2)

(b) The structure of compound A is shown in Fig. 7.1.

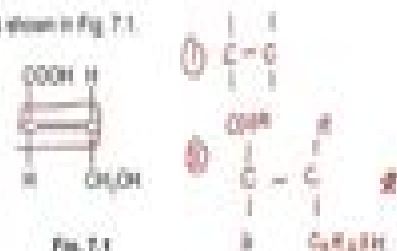


Fig. 7.1

(i) Deduce the molecular formula of compound A.

$\text{C}_4\text{H}_8\text{O}_3$ (1)

(ii) There are three functional groups in compound A.

Name the functional group of compounds that contain the following functional groups:

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Table of Contents May June 2013 Chemistry Paper 63 0620

1. Understanding the eBook May June 2013 Chemistry Paper 63 0620
 - The Rise of Digital Reading May June 2013 Chemistry Paper 63 0620
 - Advantages of eBooks Over Traditional Books
2. Identifying May June 2013 Chemistry Paper 63 0620
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an May June 2013 Chemistry Paper 63 0620
 - User-Friendly Interface
4. Exploring eBook Recommendations from May June 2013 Chemistry Paper 63 0620
 - Personalized Recommendations
 - May June 2013 Chemistry Paper 63 0620 User Reviews and Ratings
 - May June 2013 Chemistry Paper 63 0620 and Bestseller Lists
5. Accessing May June 2013 Chemistry Paper 63 0620 Free and Paid eBooks
 - May June 2013 Chemistry Paper 63 0620 Public Domain eBooks
 - May June 2013 Chemistry Paper 63 0620 eBook Subscription Services
 - May June 2013 Chemistry Paper 63 0620 Budget-Friendly Options
6. Navigating May June 2013 Chemistry Paper 63 0620 eBook Formats

- ePub, PDF, MOBI, and More
- May June 2013 Chemistry Paper 63 0620 Compatibility with Devices
- May June 2013 Chemistry Paper 63 0620 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of May June 2013 Chemistry Paper 63 0620
 - Highlighting and Note-Taking May June 2013 Chemistry Paper 63 0620
 - Interactive Elements May June 2013 Chemistry Paper 63 0620
- 8. Staying Engaged with May June 2013 Chemistry Paper 63 0620
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers May June 2013 Chemistry Paper 63 0620
- 9. Balancing eBooks and Physical Books May June 2013 Chemistry Paper 63 0620
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection May June 2013 Chemistry Paper 63 0620
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine May June 2013 Chemistry Paper 63 0620
 - Setting Reading Goals May June 2013 Chemistry Paper 63 0620
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of May June 2013 Chemistry Paper 63 0620
 - Fact-Checking eBook Content of May June 2013 Chemistry Paper 63 0620
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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