

Chapter 1
The Schrödinger Equation

1.1 (a) $\Psi_1 = 0$, (b) $\Psi_1 = 0$, (c) $\Psi_1 = 0$

1.2 (a) $E_{\text{photon}} = hf = hc/\lambda = (6.626 \times 10^{-34} \text{ J}\cdot\text{s})(3.00 \times 10^8 \text{ m/s})/(300 \times 10^{-9} \text{ m}) = 6.626 \times 10^{-19} \text{ J}$
 $= 4.14 \text{ eV} = 1.6 \times 10^{-18} \text{ J}$
 (b) $E = hf = (6.626 \times 10^{-34} \text{ J}\cdot\text{s})(3.00 \times 10^8 \text{ m/s})/(4.0 \times 10^{-9} \text{ m}) = 4.97 \times 10^{-17} \text{ J}$ and $h\nu = 3.1 \times 10^{-17} \text{ J}$

1.3 Using $E_{\text{photon}} = hf$, f given:
 $E = (6.626 \times 10^{-34} \text{ J}\cdot\text{s})(3.00 \times 10^8 \text{ m/s})/(3.00 \times 10^{-9} \text{ m}) = 6.626 \times 10^{-17} \text{ J}$

1.4 (a) $E_{\text{photon}} = hf = hc/\lambda =$
 $(6.626 \times 10^{-34} \text{ J}\cdot\text{s})(3.00 \times 10^8 \text{ m/s})/(3.00 \times 10^{-9} \text{ m}) = 6.626 \times 10^{-17} \text{ J}$
 $= 4.14 \text{ eV}$
 (b) The minimum photon energy needed to produce the photoelectric effect is $(2.75 \text{ eV})(1.602 \times 10^{-19} \text{ J/eV}) = 4.40 \times 10^{-19} \text{ J}$ and $hf = 6.626 \times 10^{-34} \text{ J}\cdot\text{s}/\lambda$ and $4.40 \times 10^{-19} \text{ J} = 6.626 \times 10^{-34} \text{ J}\cdot\text{s}/\lambda$
 (c) Since the energy needed here is smaller than the maximum E available for incident light, some photoelectrons escape and the maximum E is larger than incident E .

1.5 (a) At high frequencies, we have $h\nu \gg \phi$ and the -1 in the denominator of Planck's formula can be neglected to give Wien's formula.
 (b) The Rayleigh series for the exponential function is $e^x = 1 + x + x^2/2! + \dots$ For $x \ll 1$, we can neglect x^2 and higher powers to give $e^x \approx 1 + x$. Taking $x = hc/h\nu$, we have for Planck's formula at low frequencies:

$$\frac{e^{hc/h\nu}}{e^{hc/h\nu} - 1} \approx \frac{1 + hc/h\nu}{1 + hc/h\nu - 1} \approx \frac{1 + hc/h\nu}{hc/h\nu} = \frac{1}{hc/h\nu} + 1$$

1.6 $E = hf/\text{area} = (1.975 \times 10^{-25} \text{ J}\cdot\text{s})(3.00 \times 10^8 \text{ m/s})/(2.00 \times 10^{-2} \text{ m}) = 2.96 \times 10^{-17} \text{ J}$
 $= 1.85 \times 10^{-18} \text{ eV} = 0.00185 \text{ eV}$

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 - The Rise of Digital Reading Levine Quantum Chemistry Solutions Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Levine Quantum Chemistry Solutions Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
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 - Popular eBook Platforms
 - Features to Look for in an Levine Quantum Chemistry Solutions Manual
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 - Participating in Virtual Book Clubs
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 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Levine Quantum Chemistry Solutions Manual
 - Setting Reading Goals Levine Quantum Chemistry Solutions Manual
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Levine Quantum Chemistry Solutions Manual
 - Fact-Checking eBook Content of Levine Quantum Chemistry Solutions Manual
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

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- Interactive and Gamified eBooks

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