



Datum / Date:	NOVEMBER 2024
Tyd / Time:	2 UUR 30 MINUTE
Punte / Marks:	150 PUNTE
Graad / Grade:	T1
Wak / Subject:	LEWENSWETENSKAPPE
Vraestel / Paper:	VRAESTEL 1

Naam en Van / Name and Surname: \_\_\_\_\_

**Algemene instruksies vir die skryf van toetse | General instructions for the writing of exams:**

1. Geen vraag deur jou hele vraestel.	Quickly scan through your entire paper.
2. Lees 'n vraag 2x deur voor jy daarmee begin.	Read the question twice before you begin.
3. Maak seker jy antwoord alle vrae.	Make sure to answer ALL questions.
4. Skryf netjies en leesbaar.	Write neatly and legibly.
5. As jy te veel sukkel met 'n vraag, beweeg aan na die volgende vraag en kom terug nadat alle ander vrae beantwoord is.	If you struggle too much with a question, move on to the next one and return once you have answered all the other questions.
6. Vir langer vrae, maak seker jy beplan jou antwoord voordat jy begin skryf.	If the question calls for a longer answer, plan before you start writing.
7. Vir enige antwoord langer as 1-2 woorde, skryf altyd in volsinne.	For any answer longer than 1-2 words, always write in full sentences.
8. Maak seker elke punt verwijs terug na die sentrale vraag.	Make sure every point you make refers to the central question.
9. Gebruik die korrekte grammatika, spelling en leestekens na die beste van jou vermoë.	Use correct grammar, spelling, and punctuation to the best of your ability.
10. As jy 'n fout maak, trek dit duidelik dood met 'n enkel, nextline lyn en skryf die korrekte antwoord heen.	If you make a mistake, cross it out clearly with a single, nextline and write the correct answer instead.
11. Nummer die antwoorde korrek volgens die nummering stelsel in hierdie vraestel.	Number the answers correctly according to the numbering system used in this question paper.
12. Stel jou antwoorde korrek voor volgens die instruksies van elke vraag.	Present your answers according to the instructions for each question.

**Bele sterkte met jou eksamen! | Good luck with your exam!**

**AFDELING A**

**Vraag 1**

1.1 Verskeie opsies word as moontlike antwoorde vir die volgende vrae gegee. Kies die korrekte antwoord en skryf slegs die letter (A tot D) langs dievraagnommer ( 1.1.1 tot 1.1.10) op die FOLIO-PAPER, byvoorbeeld 1.1.11 E.

1.1.1 Groen plante wat hul eie voedsel vervaardig deur fotosintese.

- A. Katabolisme.
- B. Outotrofe.
- C. Anabolisme.
- D. Heterotrofe.

1.1.2 Vloeiende deel van chloroplast waar die lig- onafhanklike fase van fotosintese plaas vind.

- A. Lamellen.
- B. Grana.
- C. Stroma.
- D. Chlorofil.

1.1.3 Hoe ontwikkel ons 'n plant.

- A. Plaas die plant in die son.
- B. Plaas die plant in 'n donker kas.
- C. Plaas die plant in koolte.
- D. Plaas die plant in 'n deurskynende boks.

# Lewenswetenskappe Vraestel 1 November 2014 Memo

**A Loxley**



**Lewenswetenskappe Vraestel 1 November 2014 Memo:**

## The Enigmatic Realm of **Lewenswetenskappe Vraestel 1 November 2014 Memo**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Lewenswetenskappe Vraestel 1 November 2014 Memo** a literary masterpiece penned by a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those who partake in its reading experience.

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web titrant and analyte is a pair of acid and base acid base titrations are monitored by the change of ph as titration progresses let us be clear about some terminologies before we get into the discussion of titration curves titrant solution of a known concentration which is added to another solution whose concentration has to be determined

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web acid base titrations are usually used to find the amount of a known acidic or basic substance through acid base reactions the analyte titrand is the solution with an unknown molarity the reagent titrant is the solution with a known molarity that will react with the analyte

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web an acid base titration is a method of quantitative analysis for determining the concentration of an acid or base by neutralizing it using a standard solution of base or acid of known concentration a ph indicator is used to

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web an acid base titration is an experimental technique used to acquire information about a solution containing an acid or base hundreds of compounds both organic and inorganic can be determined by a titration based on their acidic or basic properties acid is titrated with a base and base is titrated with an acid

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web the titration of a weak acid with a strong base involves the direct transfer of protons from the weak acid to the hydroxide ion the reaction of the weak acid acetic acid with a strong base naoh can be seen below in the reaction the

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web as seen in the chapter on the stoichiometry of chemical reactions titrations can be used to quantitatively analyze



solutions for their acid or base concentrations in this section we will explore the underlying chemical equilibria that make acid base titrimetry a

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web in a titration a solution of known concentration the titrant is added to a solution of the substance being studied the analyte in an acid base titration the titrant is a strong base or a strong acid and the analyte is an acid or a base respectively

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