

Instructor	Course Title	Lesson Date (time)	Unit
Reem Nammour	Biology	11/12/2015 55min	introduction to chemistry
Standard	Describe the parts of the atom and types of bonds S11.C.1.1.1		
	Describe the structure of a carbon atom and how carbon atoms form macromolecules S11.C.1.1.3		
Objective		Essential Question	
Students will explore what happens to atoms when the number of subatomic particles changes through a pHet computer simulation lab.		If an atom loses an electron, what happens to it's charge? What about two or more? If an atom gains an electron, what happens to it's charge? What about two or more?	
Lesson Agenda			
Agenda/Time	5 E's	Learning Activities	
		Teacher will...	Students will ...
10 min	Engage (Quick Start/ Review)	Present students with a screen shot of an article entitled "Atom splitting in my kitchen was a hobby, man tells Swedish police". Ask students to guess why splitting an atom is considered a crime	Try to understand what it means to split an atom and why an unstable atom is dangerous. Be presented with information regarding the importance of subatomic structure and the properties of an atom.
10 min	Explore	Introduce pHet activity. Show an example of how to use the program. Present a packet with different elements and symbols in order to guide students through activity	Be expected to fully understand how the program works before they explore on their own.
30 min	Explain (Guided Practice)	Be available for any problems students may encounter. Ask guiding questions while students explore the program. Make sure students are continuing on track through the activity.	Explore pHet program while answering questions in guided lab. Will be able to understand and visualize what happens to an atom when the subatomic particle number changes. Will become familiar with the charges of an atom changing.
	Extend/ Elaborate		

# Lesson Plan Guidelines For Teachers

**L Reisser**



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