

CHEMISTRY

QUESTION PAPER

ANSWERS

CLASS 12 HSC

MAHARASTRA BOARD

20. Given data :

(i) $R = 8.314 \text{ J/K mol}$

(ii) Atomic mass $X = 12$

(iii) K_f for water = $1.86 \text{ K kg mol}^{-1}$

(iv) $1^\circ = 10000^\circ$

(v) $N_A = 6.022 \times 10^{23}$

SECTION - A

Q. 1. Select and write the correct answer for the following multiple choice type of questions :

(i) Solubility defect is NOT observed in _____



(ii) The freezing point of 1 in aqueous solution of urea, if K_f for water is $1.86 \text{ K kg mol}^{-1}$ is _____

(a) 1.86°C

(b) -1.86°C

(c) 0.186°C

(d) -0.186°C

(iii) Green layer is displayed by _____



(iv) When excess of AgNO_3 is added to a complex, one mole of AgCl is precipitated. The formula of complex is _____



(v) The value of n_F for the oxidation of 1 mole of sulphur dioxide to sulphur trioxide is _____

(a) -2

(b) -1

(c) $+1$

(d) $+2$

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