

**Chapter 10 test****Matching**

Match each item with the correct statement below.

- |                            |                                      |
|----------------------------|--------------------------------------|
| a. representative particle | d. percent composition               |
| b. mole                    | e. standard temperature and pressure |
| c. Avogadro's number       | f. empirical formula                 |

- \_\_\_\_\_ 1. the number of representative particles of a substance present in 1 mole of that substance
- \_\_\_\_\_ 2. an atom, an ion, or a molecule, depending upon the way a substance commonly exists
- \_\_\_\_\_ 3. the SI unit used to measure amount of substance
- \_\_\_\_\_ 4. 0°C and 1 atm
- \_\_\_\_\_ 5. the percent by mass of each element in a compound
- \_\_\_\_\_ 6. the smallest whole number ratio of the atoms in a compound

**Multiple Choice**

Identify the letter of the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 7. What SI unit is used to measure the number of representative particles in a substance?
- |             |           |
|-------------|-----------|
| a. kilogram | c. kelvin |
| b. ampere   | d. mole   |
- \_\_\_\_\_ 8. How many hydrogen atoms are in 5 molecules of isopropyl alcohol,  $\text{C}_3\text{H}_7\text{O}$ ?
- |                                     |                                      |
|-------------------------------------|--------------------------------------|
| a. $5 \times (6.02 \times 10^{23})$ | c. 35                                |
| b. 5                                | d. $35 \times (6.02 \times 10^{23})$ |
- \_\_\_\_\_ 9. All of the following are equal to Avogadro's number EXCEPT \_\_\_\_\_
- |  |
|--|
| a. the number of atoms of bromine in 1 mol $\text{Br}_2$     |
| b. the number of atoms of gold in 1 mol Au                   |
| c. the number of molecules of nitrogen in 1 mol $\text{N}_2$ |
| d. the number of molecules of carbon monoxide in 1 mol CO    |
- \_\_\_\_\_ 10. How many moles of tungsten atoms are in  $4.8 \times 10^{23}$  atoms of tungsten?
- |                               |                               |
|-------------------------------|-------------------------------|
| a. $8.0 \times 10^{-7}$ moles | c. $1.3 \times 10^{-4}$ moles |
| b. $8.0 \times 10^{-1}$ moles | d. $1.3 \times 10^{-2}$ moles |
- \_\_\_\_\_ 11. How many atoms are in 0.075 mol of titanium?
- |                         |                         |
|-------------------------|-------------------------|
| a. $1.2 \times 10^{23}$ | c. $6.4 \times 10^7$    |
| b. $2.2 \times 10^{24}$ | d. $4.5 \times 10^{22}$ |
- \_\_\_\_\_ 12. How many molecules are in 2.10 mol  $\text{CO}_2$ ?
- |                                    |                                     |
|------------------------------------|-------------------------------------|
| a. $2.53 \times 10^{24}$ molecules | c. $3.49 \times 10^{-24}$ molecules |
| b. $3.79 \times 10^{24}$ molecules | d. $1.26 \times 10^{24}$ molecules  |

# Modern Chemistry Chapter 10 Test B Answers

**M Planty**



## **Modern Chemistry Chapter 10 Test B Answers:**

## **Modern Chemistry Chapter 10 Test B Answers** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Modern Chemistry Chapter 10 Test B Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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