

CHAPTER 2 WORKSHEET: CHEMISTRY OF LIFE

1. Chapter 2, Section 1: Nature of Matter

- a. Matter: anything that occupies space. Ex: Desk, Human, Air, ect.
- b. Atom: Smallest part of matter. It **can not** be broken down into smaller pieces. Composed of 3 particles: Protons (+), Electrons (-), and Neutrons (n)
- c. Element: Substance that is only made of one kind of atom. Ex: Hydrogen, Oxygen, ect.
 - i. C, H, N, O, P, S
- d. Compound: Two or more different atoms or elements joined together. Ex: Salt =
- e. Covalent Bond: Two or more atoms sharing electrons to form a molecule. Covalent bonds make molecules, ex: Water, Carbon Dioxide, and Oxygen Gas
- f. Hydrogen Bond: A weak chemical attraction between **polar** molecules. Ex: Two water molecules
- g. Ionic Bond: Gain or Loss of electrons that create a molecule
- h. Ion: An atom that has gained (-) electrons or (loss) electrons (+)

2. Chapter 2, Section 2: Water and Solutions

- a. How does water help maintain homeostasis?
 - i. Heats slowly
 - ii. Retains heat
 - iii. Carries Heat away through water vapor (sweating)
 - iv. Universal solvent (anything except for a polar substance can dissolve in it)
- b. What is the difference between adhesion & cohesion?
 - i. Adhesion: different substances
 - ii. Cohesion: alike substances
- c. What is the pH range for an Acid? Base?
 - i. Acid = 0 – 6.9
 - ii. Neutral = 7
 - iii. Base = 7.1 – 14
- d. What type of substance is an example of an acid? Of a base?
 - i. Acids: Lemon, Vinegar, Milk
 - ii. Bases: Household cleaning material

3. Chapter 2, Section 3: Chemistry of Cells

- a. Carbohydrates: Organic substances, made up of carbon, hydrogen and oxygen atoms. Always in the ratio of 1:2:1. Key source of energy.
 - i. They are made up of Monosaccharide: Single sugars (glucose)
 - ii. Disaccharides are double sugars: Sucrose
 - iii. Polysaccharides: Starch and cellulose

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Answer Key

Arshad Iqbal



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alcohols reactions The chapter Atomic Structure and Theory MCQs covers topics of atom facts elements and atoms number of nucleons protons electrons and neutrons The chapter Benzene Chemical Compound MCQs covers topics of benzene arenes reaction phenol properties and reactions of phenol The chapter Carbonyl Compounds MCQs covers topics of carbonyl compounds aldehydes and ketone testing nucleophilic addition with HCN preparation of aldehydes and ketone reduction of aldehydes and ketone

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Table of Contents Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key

1. Understanding the eBook Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - The Rise of Digital Reading Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Personalized Recommendations
 - Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key User Reviews and Ratings
 - Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key and Bestseller Lists
5. Accessing Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key Free and Paid eBooks
 - Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key Public Domain eBooks
 - Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key eBook Subscription Services
 - Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key Budget-Friendly Options

6. Navigating Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key Compatibility with Devices
 - Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Highlighting and Note-Taking Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Interactive Elements Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
8. Staying Engaged with Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
9. Balancing eBooks and Physical Books Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Setting Reading Goals Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Fact-Checking eBook Content of Modern Chemistry Chapter 2 Mixed Review Worksheet Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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