

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium ($\text{H}-3$) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ + & & \\ \hline & & 63.62 \rightarrow 64 \end{array}$$

Modern Chemistry Chapter 9 Answers

Susan McMurry



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Solutions to Problems, Principles of Modern Organic Chemistry James Cason, 1967 **Solutions Manual for Modern Organic Synthesis: An Introduction** Michael Nantz, Hasan Palandoken, George S. Zweifel, 2006-08-18 This supplement includes the end of chapter problems from the main text detailed solution sets and an extra section of similar problems for grad students to study **Modern Chemistry** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

Sustainable Solutions for Modern Economies Rainer Höfer, 2009-11-20 Limited supplies of fossil fuels and concerns about global warming have created a strong desire to solve the resource issue in the age beyond petroleum This reference book from the Green Chemistry Series contains the essential areas of green chemistry and sustainability in modern economies It is the first book to outline the contribution of chemistry and of renewable chemical or biological resources to the sustainability concept and to the potential resolution of the world's energy problems It describes the current status of technical research and industrial application as well as the potential of biomass as a renewable resource for energy generation in power stations as alternative fuels and for various uses in chemistry It outlines the historical routes of the sustainability concept and specifies sustainability in metrics facts and figures The book is written by European experts from academia industry and investment banking who are world leaders in research and technology regarding sustainability alternative energies and renewable resources The sustainability aspects covered include consumer behaviour and demands lifestyles and mega trends and their impact on innovation in the industry consumer industry requirements and their impact on suppliers emerging paradigm changes in raw material demand availability sourcing and logistics the contribution of the industry to restore the life support systems of the Earth socially responsible banking and investment sustainability metrics The book highlights the potential of the different forms of renewable raw materials including natural fats and oils plant based biologically active ingredients industrial starch sucrose natural rubber wood natural fibres It also covers the actual status of biomass usage for green energy generation green transportation green chemistry and sustainable nutrition and consumer goods and it depicts the potentials of green solvents and white biotechnology for modern synthesis and manufacturing technologies The book is aimed at technical and marketing people in industry universities and institutions as well as readers in administrations and NGOs The book will also be of value to the worldwide public interested in sustainability issues and strategies as well as others interested in the practical means that are being used to reduce the environmental impact of chemical processes and products to further eco efficiency and to advance the utilization of renewable resources

Electrochemistry in Nonaqueous Solutions Kosuke Izutsu, 2009-09-22 An excellent resource for all graduate students and researchers using electrochemical techniques After introducing the reader to the fundamentals the book focuses on the latest developments in the techniques and applications in this field This second edition contains new material on environmentally friendly solvents such as room temperature ionic liquids **Modern Physical Organic Chemistry** Eric V. Anslyn, Dennis A.

Dougherty,2006 Making explicit the connections between physical organic chemistry and critical fields such as organometallic chemistry materials chemistry bioorganic chemistry and biochemistry this book escorts the reader into an area that has been thoroughly updated in recent times **Study Guide and Solutions Manual to Accompany**

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