

1. A hypothetical gas that perfectly fits all the assumptions of the kinetic-molecular theory is known as

- |                         |                   |                         |               |
|-------------------------|-------------------|-------------------------|---------------|
| <input type="radio"/> A | an imaginary gas. | <input type="radio"/> B | a real gas.   |
| <input type="radio"/> C | a perfect gas.    | <input type="radio"/> D | an ideal gas. |

2. Which of the following is an assumption of the kinetic-molecular theory of gases?

- |                         |   |                         |  |
|-------------------------|---|-------------------------|--|
| <input type="radio"/> A | Gases consist of closely spaced particles.      | <input type="radio"/> B | Collisions between gas particles are inelastic.                                      |
| <input type="radio"/> C | Gas particles move around in an orderly manner. | <input type="radio"/> D | The temperature of a gas depends on the average kinetic energy of the gas particles. |

3. A liquid can be described as a form of matter that

- |                         |                       |                         |                        |
|-------------------------|-----------------------|-------------------------|------------------------|
| <input type="radio"/> A | has a definite shape. | <input type="radio"/> B | has a definite volume. |
| <input type="radio"/> C | cannot diffuse.       | <input type="radio"/> D | has low density.       |

4. Diffusion is much slower in liquids than in gases because

- |                         |                                       |                         |                                      |
|-------------------------|---------------------------------------|-------------------------|--------------------------------------|
| <input type="radio"/> A | liquids have surface tension.         | <input type="radio"/> B | liquids exist at lower temperatures. |
| <input type="radio"/> C | liquid particles are closer together. | <input type="radio"/> D | liquid particles weigh more.         |

5. The three states of matter of a particular substance, in order of the strength of intermolecular forces from least to greatest, are

- |                         |                    |                         |                    |
|-------------------------|--------------------|-------------------------|--------------------|
| <input type="radio"/> A | solid, liquid, gas | <input type="radio"/> B | gas, liquid, solid |
| <input type="radio"/> C | liquid, solid, gas | <input type="radio"/> D | gas, solid, liquid |

6. Which is an example of a noncrystalline solid?

- |                         |     |                         |       |
|-------------------------|-----|-------------------------|-------|
| <input type="radio"/> A | ice | <input type="radio"/> B | glass |
|-------------------------|-----|-------------------------|-------|

# Modern Chemistry Quiz Ch 9

**John Elling Tufte**



## **Modern Chemistry Quiz Ch 9:**

*Instructor's Manual and Test Bank to Accompany Basic Concepts of Chemistry* Malone, 2000-04-18      *Survey of Science History & Concepts Parent Lesson Plan*, 2013-08-01

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Semester 2 Biology and Chemistry The field of biology focuses on living things from the smallest microscopic protozoa to the largest mammal In this book you will read and explore the life of plants insects spiders and other arachnids life in water reptiles birds and mammals highlighting God s amazing creation You will learn about biological classification how seeds spread around the world long term storage of energy how biologists learned how the stomach digested food the plant that gave George de Mestral the idea of Velcro and so much more For most of history biologists used the visible appearance of plants or animals to classify them They grouped plants or animals with similar looking features into families Starting in the 1990 s biologists have extracted DNA and RNA from cells as a guide to how plants or animals should be grouped Like visual structures these reveal the underlying design of creation *Exploring the World of Biology* is a fascinating look at life from the smallest proteins and spores to the complex life systems of humans and animals Chemistry is an amazing branch of science that affects us every day yet few people realize it or even give it much thought Without chemistry there would be nothing made of plastic there would be no rubber tires no tin cans no televisions no microwave ovens or something as simple as wax paper This book presents an exciting and intriguing tour through the realm of chemistry as each chapter unfolds with facts and stories about the discoveries of discoverers Find out

why pure gold is not used for jewelry or coins Join Humphry Davy as he made many chemical discoveries and learn how they shortened his life See how people in the 1870s could jump over the top of the Washington Monument Exploring the World of Chemistry brings science to life and is a wonderful learning tool with many illustrations and biographical information

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