

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

## Assessment

## Chapter Test A

## Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
  - a slippery feel
  - the ability to conduct an electric current
  - reactivity with metals
- \_\_\_\_\_ 2. Which is an example of a weak acid?
- HCl
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_3\text{PO}_4$
  - $\text{HNO}_3$
- \_\_\_\_\_ 3. Acetic acid is found in
- vinegar.
  - the stomach.
  - antacids.
  - oranges.
- \_\_\_\_\_ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
  - tastes bitter.
  - dissociates to form  $\text{OH}^-$  ions.
  - All of the above
- \_\_\_\_\_ 5. Which of the following is a Brønsted-Lowry base?
- HCl
  - $\text{HCO}_3^-$
  - $\text{H}_3\text{O}^+$
  - $\text{H}_3\text{PO}_4$
- \_\_\_\_\_ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(aq) + \text{CH}_3\text{NH}_2(aq) \rightleftharpoons \text{HC}_2\text{O}_4^-(aq) + \text{CH}_3\text{NH}_3^+(aq),$$
- which of these is a conjugate acid-base pair?
- $\text{H}_2\text{C}_2\text{O}_4(aq)$  and  $\text{CH}_3\text{NH}_2(aq)$
  - $\text{H}_2\text{C}_2\text{O}_4(aq)$  and  $\text{CH}_3\text{NH}_3^+(aq)$
  - $\text{CH}_3\text{NH}_2(aq)$  and  $\text{CH}_3\text{NH}_3^+(aq)$
  - $\text{HC}_2\text{O}_4^-(aq)$  and  $\text{CH}_3\text{NH}_3^+(aq)$

# Modern Chemistry Assessment Chapter Test Reaction Energy

**Lei Huang**



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## **Table of Contents Modern Chemistry Assessment Chapter Test Reaction Energy**

1. Understanding the eBook Modern Chemistry Assessment Chapter Test Reaction Energy
  - The Rise of Digital Reading Modern Chemistry Assessment Chapter Test Reaction Energy
  - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Assessment Chapter Test Reaction Energy
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in a Modern Chemistry Assessment Chapter Test Reaction Energy
  - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Assessment Chapter Test Reaction Energy
  - Personalized Recommendations
  - Modern Chemistry Assessment Chapter Test Reaction Energy User Reviews and Ratings
  - Modern Chemistry Assessment Chapter Test Reaction Energy and Bestseller Lists
5. Accessing Modern Chemistry Assessment Chapter Test Reaction Energy Free and Paid eBooks
  - Modern Chemistry Assessment Chapter Test Reaction Energy Public Domain eBooks
  - Modern Chemistry Assessment Chapter Test Reaction Energy eBook Subscription Services
  - Modern Chemistry Assessment Chapter Test Reaction Energy Budget-Friendly Options
6. Navigating Modern Chemistry Assessment Chapter Test Reaction Energy eBook Formats
  - ePub, PDF, MOBI, and More
  - Modern Chemistry Assessment Chapter Test Reaction Energy Compatibility with Devices
  - Modern Chemistry Assessment Chapter Test Reaction Energy Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Modern Chemistry Assessment Chapter Test Reaction Energy
  - Highlighting and Note-Taking Modern Chemistry Assessment Chapter Test Reaction Energy
  - Interactive Elements Modern Chemistry Assessment Chapter Test Reaction Energy

8. Staying Engaged with Modern Chemistry Assessment Chapter Test Reaction Energy
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Modern Chemistry Assessment Chapter Test Reaction Energy
9. Balancing eBooks and Physical Books Modern Chemistry Assessment Chapter Test Reaction Energy
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Modern Chemistry Assessment Chapter Test Reaction Energy
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Modern Chemistry Assessment Chapter Test Reaction Energy
  - Setting Reading Goals Modern Chemistry Assessment Chapter Test Reaction Energy
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modern Chemistry Assessment Chapter Test Reaction Energy
  - Fact-Checking eBook Content of Modern Chemistry Assessment Chapter Test Reaction Energy
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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