

## Chapter Test B

### Chapter: Chemical Equations and Reactions

Match a term or name (probably written in bold in the text or phrase describing completed tests) to each of the test answers with questions.

- \_\_\_\_\_ 1. The production of a slightly soluble substance is a double displacement reaction. Which is the formula of a \_\_\_\_\_  
a. gas  
**b. precipitate**  
c. colorless solution  
d. halogen
- \_\_\_\_\_ 2. The balance chemical equation is experimentally useful for the \_\_\_\_\_  
**a. prediction**  
b. calculation  
c. structure of the products  
d. number of products
- \_\_\_\_\_ 3. In chemical equation, the coefficient of  $\text{H}_2$  in the following equation is \_\_\_\_\_  
**a. seven**  
**b. multiplied by seven**  
c. one-half  
d. multiplied
- \_\_\_\_\_ 4. The following test procedures are used to determine if the \_\_\_\_\_  
a. chemical formula of the reactants is the same as the chemical formula of the products  
b. temperature decreases  
c. reaction takes place  
**d. complete formation of the product for the reaction between the reactants**
- \_\_\_\_\_ 5. The coefficient of the chemical equation \_\_\_\_\_  
**a. indicates the number of moles of each substance that react**  
b. shows the number of grams of each substance that react  
c. shows the number of moles of the products  
d. shows the number of moles of each atom
- \_\_\_\_\_ 6. If species A is known to be a reactant in the reaction  $\text{A} + \text{B} \rightarrow \text{C}$ , then \_\_\_\_\_  
a. A will appear on the left side of the equation  
**b. A will appear on the right side of the equation**  
c. A will appear on both sides of the equation  
d. A will appear with a coefficient

# Modern Chemistry Ch 11 Test B Answers

**Zumdahl, Steven S. Zumdahl**



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