

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

61.5 yr

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} \boxed{12.5 \text{ mg}}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ \hline & + & \\ & 63.62 & \rightarrow 64 \end{array}$$

Modern Chemistry Chapter Section 4 Review Answers

JE Gale



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Study Guide to accompany Basic Concepts of Chemistry, 7th Edition** Leo J. Malone, 2003-02-20 Work more effectively and gauge your progress along the way This Study Guide that is designed to accompany Malone's Chemistry 7th Edition includes chapter summaries new terms self tests answers to self tests and solutions to selected problems This easy to read introduction presents chemistry as a living relevant science Chemistry 7th Edition encourages critical thinking and helps readers overcome the math difficulties that often prevent them from developing a full understanding of the subject This new seventh edition builds on its core strengths of pedagogy driving the connections between ideas mathematics in context not just an appendix and an extensive problem solving emphasis with an updated design and more molecular art In addition the seventh edition expands its applications and online options One of the briefest books in the market it still provides sufficient depth for the basic concepts of chemistry Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000 The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems The Well-Trained Mind Susan Wise Bauer, Jessie Wise, 2024-04-02 The classic handbook on home schooling updated for a new generation of parents and students Is your child getting lost in the system becoming bored losing his or her natural eagerness to learn If so it may be time to take charge of your child's education by doing it yourself The Well Trained Mind will instruct you step by step on how to give your child an academically rigorous comprehensive education from preschool through high school one that will train him or her to read to think to understand to be well rounded and curious about learning Veteran home educator Susan Wise Bauer outlines the classical pattern of education called the trivium which organizes learning around the maturing capacity of the child's mind and comprises three stages the elementary school grammar stage when the building blocks of information are absorbed through memorization and rules the middle school logic stage in which the student begins to think more analytically and the high school rhetoric stage where the student learns to write and speak with force and originality Using this theory as your model you'll be able to instruct your child whether full time or as a supplement to classroom education in all levels of reading writing history geography mathematics science foreign languages rhetoric logic art and music regardless of your own aptitude in those subjects A new optional Resource Recommendations Portal provides subscribers with curated lists of the best curricula for every grade level and learning style Thousands of parents have already used the methods described in The Well Trained Mind to create a truly

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The School Science Review ,1940 **Green Chemistry and Computational Chemistry** Liliana L. Mammino,2021-11-17 Green chemistry already draws on many techniques and approaches developed by theoretical chemists whilst simultaneously revealing a whole range of interesting new challenges for theoretical chemists to explore Highlighting how work at the intersection of these fields has already produced beneficial results Green Chemistry and Computational Chemistry Shared Lessons in Sustainability is a practical informative guide to combining green and theoretical chemistry principles and approaches in the development of more sustainable practices Beginning with an introduction to both theoretical chemistry and green chemistry the book goes on to explore current approaches being taken by theoretical chemists to address green and sustainable chemistry issues before moving on to highlight ways in which green chemists are employing the knowledge and techniques of theoretical chemistry to help in developing greener processes The future possibilities for theoretical chemistry in addressing sustainability issues are discussed before a selection of case studies provides good insight into how these interactions and approaches have been successfully used in practice Highlights the benefits of green and theoretical chemistry groups working together to tackle sustainability issues across both academia and industry Supports readers in easily selecting the most appropriate path through the book for their own needs Presents a range of examples examining the practical implications and outcomes of interdisciplinary approaches Predicting Chemical Toxicity and Fate Mark T.D. Cronin,2004-05-10 Quantitative Structure Activity Relationships QSARs are increasingly used to predict the harmful effects of chemicals to humans and the environment The increased use of these methods in a variety of areas academic industrial regulatory results from a realization that very little toxicological or fate data is available on the vast amount of chemicals to which humans and the environment are exposed Predicting Chemical

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Introduction to Modern Chemistry: Student Manual Eugene Meyer, 1979 *Aromatic Chemistry* P.W.G. Smith, A.R. Tatchell, 2016-06-27 Organic Chemistry for General Degree Students Volume 2 Aromatic Chemistry presents the fundamental aspects of aromatic chemistry. This book explores the systematic study in the first instance of the chemistry of functional groups based on their structural characteristics in aliphatic systems. Organized into 11 chapters, this volume begins with an overview of the properties of the aromatic nucleus followed by a consideration of the manner in which interaction with the aromatic system may alter the reactivity of functional groups. Other chapters explain the two distinctly different classes of aromatic halogen compounds. This text discusses as well the properties of an aromatic amine which is characterized by having at least one aromatic system attached to the nitrogen atom and may be further categorized as primary, secondary or tertiary according to the degree of substitution of the nitrogen. The final chapter examines the classical structure for thiophene. This book is a valuable resource for organic chemists and students.

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