

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium ($\text{H}-3$) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 & \times & .692 = 43.596 \rightarrow 44 \\ 65 & \times & .308 = 20.02 \rightarrow 20. \\ & + & \\ & \hline & & 63.62 \rightarrow 64 \end{array}$$

Modern Chemistry Review Answers Section 2

Reginald Robert Hawkins



Modern Chemistry Review Answers Section 2:

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Organic Synthesis* Michael Smith, 2011-07-12 The first two chapters provide an introduction to functional groups these are followed by chapters reviewing basic organic transformations e g oxidation reduction The book then looks at carbon carbon bond formation reactions and ways to disconnect a bigger molecule into simpler building blocks Most chapters include an extensive list of questions to test the reader s understanding There is also a new chapter outlining full retrosynthetic analyses of complex molecules which highlights common problems made by scientists **Saturday Review** ,1893 **The Academy and Literature** ,1890 *The Saturday Review of Politics, Literature, Science and Art* ,1876 **Solutions Guide for Introductory Chemistry** James F. Hall, 1996 **The Cambridge Review** ,1888 **The University correspondent and University correspondence college magazine (and The Educational review).** ,1890 **The Publishers Weekly** ,1917 **Brewers' Journal and Hop and Malt Trades' Review** ,1915 *Educational Times* ,1890 **The Critical Review or Annals of Literature, 1756-1763 Vol 1** James G Basker, 2024-10-28 The Critical Review reflects the political scientific and literary debate of the times The journal was edited for its first seven years by Tobias Smollett and reflected the slashing combative style and intellectual range of its editor This 16 volume set reproduces this journal **Technical Book Review Index** ,1972 Academy, with which are Incorporated Literature and the English Review ,1891 **Fossil Energy Update** ,1986 *Practical Druggist and Pharmaceutical Review of Reviews* ,1915 **Scientific, Medical** Reginald Robert Hawkins, 1950 **Scientific, Medical and Technical Books Published in the United States of America, 1930-1944 ...** Reginald Robert Hawkins, 1953 **High School Biology Review** Kim Magloire, Princeton Review (Firm), 1998-03 If trudging through your textbook to study and complete homework assignments has become a frustrating grind then get ready for a smooth ride to higher test scores and outstanding grades with The Princeton Review s High School Biology Review We tell it to you straight thoroughly explaining the important topics you ll need to understand to prepare for quizzes and tests complete homework assignments effectively and earn higher grades We ve carefully examined biology textbooks just like yours to make sure that this book includes all the material essential to a thorough review In this guide we cover The Chemistry of Life Life Functions Biodiversity The Cell Reproduction Ecology Cellular Respiration Genetics Laboratory Skills Photosynthesis Modern Genetics Plants Evolution And since practicing your test taking skills is just as important to getting good grades as knowing the material we include two practice exams that feature the types of questions and problems that appear on in class tests *Leather Trades' Review* ,1903

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