

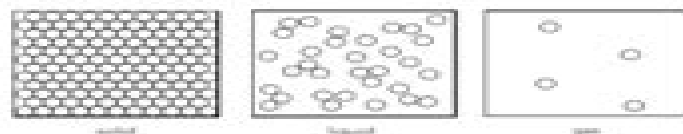
# > Workbook answers

## Chapter 1

### Exercise 1.1

- 1 A solid has a fixed **mass** and **volume / shape**.  
A liquid has a fixed **mass** but its **shape** changes to that of the container in which it is placed.  
A gas has no fixed **shape** or **volume**. A gas completely fills the container that it is in.

2



- 3 a B  
b C  
c A  
d D
- 4 A freezing (solidification)  
B melting  
C condensation  
D evaporation or boiling
- 5 a radon  
b radon and nitrogen  
c nitrogen  
d cobalt  
e The sample of ethanoic acid is impure. The presence of impurities raises the boiling point of a substance.

### Exercise 1.2

6



- 7 The student would need to use an oil bath (in place of the water bath) so that the higher temperature could be reached.
- 8 The substance is freezing (solidifying) / turning from liquid to solid.
- 9 The temperature stays constant because energy is being released as the substance solidifies / the molecules are giving out heat as they stop moving from place to place and become organised in a structured lattice arrangement / in the solid the molecules can only vibrate about fixed points / the heat released by the formation of new interactions (forces) between the particles keeps the temperature constant until all the substance is solid.

# Modern Chemistry Workbook Answers Chapter 11

**RD Boyd**



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### **Table of Contents Modern Chemistry Workbook Answers Chapter 11**

1. Understanding the eBook Modern Chemistry Workbook Answers Chapter 11
  - The Rise of Digital Reading Modern Chemistry Workbook Answers Chapter 11
  - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Workbook Answers Chapter 11
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in a Modern Chemistry Workbook Answers Chapter 11
  - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Workbook Answers Chapter 11
  - Personalized Recommendations
  - Modern Chemistry Workbook Answers Chapter 11 User Reviews and Ratings

- Modern Chemistry Workbook Answers Chapter 11 and Bestseller Lists
- 5. Accessing Modern Chemistry Workbook Answers Chapter 11 Free and Paid eBooks
  - Modern Chemistry Workbook Answers Chapter 11 Public Domain eBooks
  - Modern Chemistry Workbook Answers Chapter 11 eBook Subscription Services
  - Modern Chemistry Workbook Answers Chapter 11 Budget-Friendly Options
- 6. Navigating Modern Chemistry Workbook Answers Chapter 11 eBook Formats
  - ePub, PDF, MOBI, and More
  - Modern Chemistry Workbook Answers Chapter 11 Compatibility with Devices
  - Modern Chemistry Workbook Answers Chapter 11 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Modern Chemistry Workbook Answers Chapter 11
  - Highlighting and Note-Taking Modern Chemistry Workbook Answers Chapter 11
  - Interactive Elements Modern Chemistry Workbook Answers Chapter 11
- 8. Staying Engaged with Modern Chemistry Workbook Answers Chapter 11
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Modern Chemistry Workbook Answers Chapter 11
- 9. Balancing eBooks and Physical Books Modern Chemistry Workbook Answers Chapter 11
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Modern Chemistry Workbook Answers Chapter 11
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Modern Chemistry Workbook Answers Chapter 11
  - Setting Reading Goals Modern Chemistry Workbook Answers Chapter 11
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modern Chemistry Workbook Answers Chapter 11
  - Fact-Checking eBook Content of Modern Chemistry Workbook Answers Chapter 11
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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