

## Similarity Worksheet

Name \_\_\_\_\_ Date \_\_\_\_\_

Solve the following problems. Show all work and include the units in both the setup and the final answer.

1. How many grams of  $\text{NaCl}$  are needed to prepare 1.00 liter of a 1.00 molar solution?
2. What mass of  $\text{Na}_2\text{CO}_3$  is needed to prepare 0.50 liter of a 2.00 molar solution?
3. The molar mass of  $\text{NaCl}$  is 58.44 g/mol. What mass of sodium is required?
4. What mass of iron(III)sulfate is needed to make 500 mL of a 0.25 M solution?
5. How many grams of iron sulfate are needed to prepare 500 mL of a 0.500 M solution?
6. How many grams of sodium are needed to prepare 1.00 L of a 0.250 M solution of  $\text{NaOH}$ ?
7. Find the mass of potassium sulfate needed to prepare 0.500 L of a 0.750 M solution of this salt.
8. What mass of  $\text{CaCl}_2$  is needed to prepare 0.500 L of a 0.750 M solution?
9. If 1.50 grams of  $\text{CaSO}_4$  are needed to prepare 1.00 mL of solution, what will be the molarity of the solution?
10. What is the molarity of a solution of 500 mL of solution contains 9.00 g of  $\text{Na}_2\text{CO}_3$ ?
11. What is the molarity of a solution that contains 0.50 g of sodium, used to find mL of solution?
12. What is the molarity of a solution that contains 0.00 g of  $\text{NaCl}$  in 1.00 mL of solution?
13. What is the molarity of three mL solution of  $\text{NaOH}$  in 0.000 grams of solution?
14. Find the concentration in molarity of a solution that contains 0.75 g sodium  $\text{CaSO}_4$  in a solution of 0.750 L.
15. What volume of solution would be prepared if 0.00 g  $\text{CaCl}_2$  would be used to make a 0.000 M solution?

# Molarity Chemistry Answers Pg 68

**Peter E. Demmin**



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