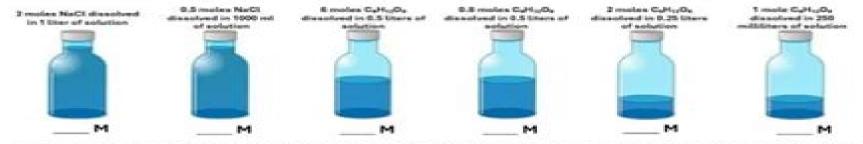
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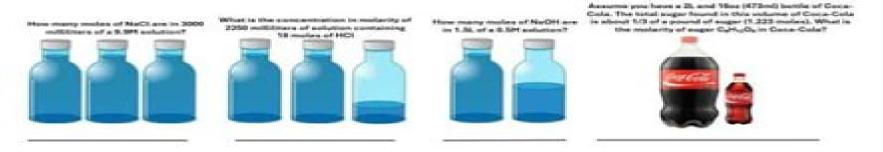
- 1. Molarity is a type of \_\_\_\_\_\_\_ similar to %, X mg/ml. (see: weet-more with a
- 2. Molarity is defined as # of dissolved within one of solution.
- 3. M is the unit for molarity, but the unit for molarity can also be written as:
- 4. Analyze the images below and identify the molarity of the solutions pictured.



5. Analyze the images below and identify how many total moles of the compound are present



Analyze the images below and answer the prompts accordingly. Be sure to include the
correct units and identity of the solution on question. Ex: 2 moles CH<sub>4</sub> or 3M MgCl



# **Molarity Concept Review**

**Karin Nielsen-Saines** 

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