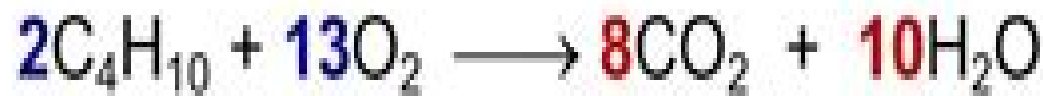


Mole Ratios Between Molecules:



Mole ratio of A:B:C:D = a:b:c:d



2 moles of C_4H_{10} will react with 13 moles of O_2 .

1 mole of C_4H_{10} will react with how many moles of O_2 ?

$$1 \text{ mole } C_4H_{10} \times \frac{13 \text{ moles } O_2}{2 \text{ moles } C_4H_{10}} = 6.5 \text{ moles } O_2$$

Mole Ratio Of A Chemical Reaction Packet

Sabine Zange



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