

(1) Given this equation: N2 + 3 H2 ---> 2 NH3, write the following molar ratios:

② Given the following equation: 8 H<sub>2</sub> + S<sub>8</sub> --> 8 H<sub>2</sub>S, write the following molar ratios:

(3) Answer the following questions for this equation: 2 H<sub>2</sub> + O<sub>2</sub> ---> 2 H<sub>2</sub>O

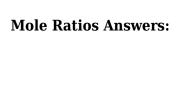
b) Suppose you had 20 moles of 
$$H_2$$
 on hand and plenty of  $O_2$ , how many moles of  $H_2O$  could you make? 20 moles  $H_2$  2  $H_2O$  = 20 moles  $H_2O$ 

4) Use this equation: N2 + 3 H2 --> 2 NH3, for the following problems

d) If 0.600 moles of NH3 were produced, how many moles of H2 are required?

# **Mole Ratios Answers**

Ying Liu



#### Mole Ratios Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Mole Ratios Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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