

Chapter 11

Molecular Composition of Gases

11-1 Volume-Mass Relationships of Gases

Avogadro's Law, STP, molar volume

11-2 The Ideal Gas Law

$PV=nRT$ ($R = 0.0821 \text{ (L}\cdot\text{atm)/(mol}\cdot\text{K)}$), finding molar mass and density

11-3 Stoichiometry of Gases

11-4 Effusion and Diffusion

Graham's law $(\sqrt{M_B})/(\sqrt{M_A})$ (assuming constant T and P)

Molecular Composition Of Gases 11 2 Answers

IM Harris



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