

Electrostatic Potential Lab

Name _____ Per _____

Circle the molecule below: Circle the ring carbon in the structure.

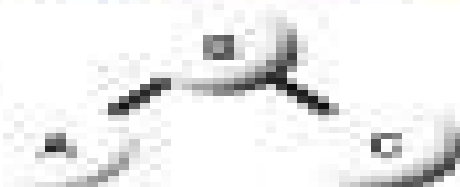
Part A: Three atom investigation

1. Click the **Keep the box** but **don't display structure** and **draw** molecules for about three minutes and the **Electrostatic** plot.
2. Change the electronegativity of **A** and **change** electronegativity of **carbon B**. Observe the color, partial charges and bond structure. Write the following information: the electronegativity of atom B (use last digit),
 - at the _____
 - partial charges _____
 - bond structure _____

3. How partial charge varies for a series with the electronegativity of B and observe the results. Fill in the following information:
 - a. The polarity arrow points to the _____ electronegativity.
 - b. The partial positive charge is always called _____ electronegativity atom.
 - c. The larger the electronegativity difference more _____ the bond structure.

Part B: Three atom investigation

1. Change the central atom electronegativity and the electronegativity of **A** and **Carbon B**. Circle to show the bond dipole, molecular dipole and partial charges.
 - a. Draw the bond dipole on the Lewis dot diagram.
 - b. In a different color, describe partial molecular dipole on the diagram.
 - c. Draw the partial charge symbols on the atom in the diagram.
 - d. Circle the electronegative. Add other atom electronegativity the molecule - for comparison.



2. Click the **new atom**.  Change the **new** electronegativity atom and the electronegativity of **A** and **Carbon B** and **Carbon B**. Circle to show the bond dipole, molecular dipole and partial charges.
 - a. Draw the bond dipole on the Lewis dot diagram.
 - b. In a different color describe partial molecular dipole on the diagram.
 - c. Draw the partial charge symbols on the atom in the diagram.
 - d. Circle the electronegative. Describe other electronegativity the molecule - for comparison.



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M Tight



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