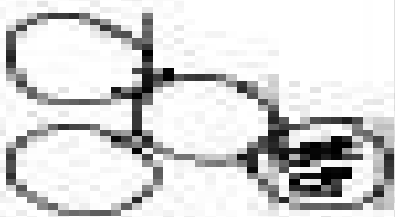


ChemThink: Molecular Shapes

Objective:
Unit 8
Electrons
Val
Electron

Understanding that as the number of molecules that have lone pairs on the central atom of molecules, the shape of the molecule will be affected, and it affects the shape of the molecule.

Number of lone pairs on central atom	Diagram (see example)	Bond Angles	Shape	Ex. Lewis Dot representation
0		180°	Linear	$H-H$
1		120°	Trigonal Planar	
2		120°	Bent	

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