# MOLES, GRAMS AND PARTICLES QUESTIONS AND ANSWERS

What is the mass, in grams, of 6.11 mol of sulfur trioxide?

A) 80.1 g B) 48.6 g C) 391 g D) 489 g E) 801 g - answer--489 g

One mol of particles of any substance contains how many particles?

A) 10° B) 3 × 10<sup>-10</sup> C) 3 × 10<sup>10</sup> D) 6.022 × 10<sup>23</sup> E) 6.022 × 10<sup>-23</sup> - answer--6.022 × 10<sup>23</sup>

Avogadro's number is the number of

A) particles in 1 mol of a substance
B) amu in 1 mol of a substance
C) grams in 1 mol of a substance
D) moles in 6.022 × 10<sup>23</sup> grams of an element
E) moles in 6.022 × 10<sup>23</sup> amu of an element - answer--particles in 1 mol of a substance

What is the mass of 0.00100 mol of caffeine, C<sub>8</sub>H<sub>10</sub>N<sub>4</sub>O<sub>3</sub>?

A) 194 g B) 87.0 g C) 0.194 g D) 0.870 g E) 1.94 g - answer--0.194 g

What is the mass of 2.00 mol of bromine, a liquid element?

A) 16.0 g B) 79.9 g C) 70.0 g D) 160. g E) 320. g - answer--320. g

0.100 mol of lithium has a mass of

# **Moles And Particles Answers**

**JA Banks** 



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