

1.  $2H_2 + O_2 \rightarrow 2H_2O$
2.  $H_2 + Cl_2 \rightarrow 2HCl$
3.  $PbSO_4 + Zn \rightarrow ZnSO_4 + Pb$
4.  $4B + 3O_2 \rightarrow 2B_2O_3$
5.  $2Al + 6HCl \rightarrow 2AlCl_3 + 3H_2$
6.  $N_2 + O_2 \rightarrow 2NO$
7.  $P_4 + 3O_2 \rightarrow P_4O_6$
8.  $4NH_3 + 5O_2 \rightarrow 6H_2O + 4NO$
9.  $Mg + 2HNO_3 \rightarrow Mg(NO_3)_2 + H_2$
10.  $Hg + S \rightarrow HgS$
11.  $CaO + SO_2 \rightarrow CaSO_3$
12.  $MgO + H_2SO_4 \rightarrow MgSO_4 + H_2O$
13.  $Zn + 2H_2O \rightarrow Zn(OH)_2 + H_2$
14.  $MgSO_4 + Na_2CO_3 \rightarrow MgCO_3 + Na_2SO_4$
15.  $Cl_2 + 2F_2 \rightarrow 2ClF_3$
16.  $Xe + 3F_2 \rightarrow XeF_6$
17.  $Al_2O_3 \cdot 2H_2O \rightarrow Al_2O_3 + 2H_2O$
18.  $2PbS + 3O_2 \rightarrow 2PbO + 2SO_2$
19.  $2HgO \rightarrow 2Hg + O_2$
20.  $Cr_2O_3 + 2Al \rightarrow Al_2O_3 + 2Cr$

# Lab 20 Balanced Chemical Equations

**J Elliott**



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