

CHAPTER 22 REVIEW***Organic Chemistry*****MIXED REVIEW****SHORT ANSWER** Answer the following questions in the space provided.

1. _____ A saturated organic compound
- (a) contains all single bonds.
 - (b) contains at least one double or triple bond.
 - (c) contains only carbon and hydrogen atoms.
 - (d) is quite soluble in water.
2. Arrange the following in order of increasing boiling point:
- _____ a. ethane
 - _____ b. pentane
 - _____ c. heptadecane
3. Recall that isomers in organic chemistry have identical molecular formulas but different structures and IUPAC names.
- _____ a. Two isomers must have the same molar mass. True or False?
 - _____ b. Two isomers must have the same boiling point. True or False?
4. Explain why hydrocarbons with only single bonds cannot form geometric isomers.

5. Write the IUPAC name for the following structural formulas:



Modern Chemistry Answers Ch 21 Mixed Review

John Bruce Wallace



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Modern Powder Diffraction* David L. Bish, Jeffrey E. Post, 2018-12-17 Volume 20 of Reviews in Mineralogy attempted to 1 provide examples illustrating the state of the art in powder diffraction with emphasis on applications to geological materials 2 describe how to obtain high quality powder diffraction data and 3 show how to extract maximum information from available data In particular the nonambient experiments are examples of some of the new and exciting areas of study using powder diffraction and the interested reader is directed to the rapidly growing number of published papers on these subjects Powder diffraction has evolved to a point where considerable information can be obtained from μg sized samples where detection limits are in the hundreds of ppm range and where useful data can be obtained in milliseconds to microseconds We hope that the information in this volume will increase the reader's access to the considerable amount of information contained in typical diffraction data

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