

APPENDIX D*Problem Bank*

- 1. a.** Given: 5.2 cm
Unknown: length in millimeters
$$5.2 \text{ cm} \times \frac{10 \text{ mm}}{1 \text{ cm}} = 52 \text{ mm}$$
- b.** Given: 0.049 kg
Unknown: mass in grams
$$0.049 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} = 49 \text{ g}$$
- c.** Given: 1.60 mL
Unknown: volume in microliters
$$1.60 \text{ mL} \times \frac{1000 \text{ }\mu\text{L}}{1 \text{ mL}} = 1600 \text{ }\mu\text{L}$$
- d.** Given: 0.0025 g
Unknown: mass in micrograms
$$0.0025 \text{ g} \times \frac{1000000 \text{ }\mu\text{g}}{1 \text{ g}} = 2500 \text{ }\mu\text{g}$$
- e.** Given: 0.020 kg
Unknown: mass in milligrams
$$0.020 \text{ kg} \times \frac{1000000 \text{ mg}}{1 \text{ kg}} = 20000 \text{ mg}$$
- f.** Given: 3 kL
Unknown: volume in liters
$$3 \text{ kL} \times \frac{1000 \text{ L}}{1 \text{ kL}} = 3000 \text{ L}$$
-
- 2. a.** Given: 150 mg
Unknown: mass in grams
$$150 \text{ mg} \times \frac{1 \text{ g}}{1000 \text{ mg}} = 0.15 \text{ g}$$
- b.** Given: 2500 mL
Unknown: volume in liters
$$2500 \text{ mL} \times \frac{1 \text{ L}}{1000 \text{ mL}} = 2.5 \text{ L}$$
- c.** Given: 0.5 g
Unknown: mass in kilograms
$$0.5 \text{ g} \times \frac{1 \text{ kg}}{1000 \text{ g}} = 0.0005 \text{ kg}$$
- d.** Given: 55 L
Unknown: volume in kiloliters
$$55 \text{ L} \times \frac{1 \text{ kL}}{1000 \text{ L}} = 0.055 \text{ kL}$$

Modern Chemistry Appendix D Answers

Bill W. Tillery



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Principles of Modern Chemistry David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom *Study and Problem Solving Guide to Accompany Principles of Modern Chemistry*, Oxtoby/Nachtrieb Wade A. Freeman, 1986 *Prentice Hall Chemistry*, 2000 Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2012-04-13 Basic Concepts of Chemistry was originally written over thirty years ago to address the needs of general chemistry students with little or no background in chemistry Over time the text has evolved beyond purposes solely aimed at a preparatory chemistry course For some preparatory chemistry students a main sequence in general chemistry may follow but for others a semester of organic and biochemistry may follow Other students enroll to simply satisfy a basic science or chemistry requirement The text was written with a level and functionality designed to accommodate the needs of each of these varied groups of students Basic Concepts of Chemistry was designed with a flexibility that allows instructors to emphasize or omit certain clearly delineated sections The mission of the text has evolved in response to the increased diversity of students and the emphasis on outcomes assessment **Modern Chemistry** Nicholas D. Tzimopoulos, 1993 **A Guide to Modern Chemistry** John F. Baxter, Jay A. Young, 1959 Minutes of the General Medical Council, and of Its Various Committees and Branch Councils General Medical Council (Great Britain), 1904 *Chemistry* William Franklin Kieffer, 1971 For college or university students majoring in other than the natural sciences **Chemistry in the Modern World** Frank L. Wiseman, 1985 **Introduction to Physics and Chemistry** Bill W. Tillery, 1992 Education Outlook, 1914 *The Educational Times and Education Outlook*, 1924 *Educational Times*, 1896 **The Educational Times, and Journal of the College of Preceptors**, 1892 *The Athenaeum*, 1896 **The New Orleans Medical and Surgical Journal**, 1928 *Catalogue of Books Recommended by the Ontario Department of Education for Libraries of Collegiate Institutes, High Schools, and Continuation Schools* Ontario. Department of Education, 1918 **The Academy and Literature**, 1893 *Academy; a Weekly Review of Literature, Learning, Science and Art*, 1896 The Poetical gazette the official organ of the Poetry society and a review of poetical affairs nos 4 7 issued as supplements to the Academy v 79 Oct 15

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