

## Chemistry Chapter 5 Exam

### Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to
  - a. Mendeleev.
  - b. Moseley.
  - c. Bohr.
  - d. Ramsay.
- \_\_\_\_\_ 2. Mendeleev left spaces in his periodic table and predicted several elements and their
  - a. atomic numbers.
  - b. colors.
  - c. properties.
  - d. radioactivity.
- \_\_\_\_\_ 3. Mendeleev attempted to organize the chemical elements based on their
  - a. symbols.
  - b. properties.
  - c. atomic numbers.
  - d. electron configurations.
- \_\_\_\_\_ 4. Mendeleev noticed that properties of elements usually repeated at regular intervals when the elements were arranged in order of increasing
  - a. atomic number.
  - b. density.
  - c. reactivity.
  - d. atomic mass.
- \_\_\_\_\_ 5. Mendeleev is credited with developing the first successful
  - a. periodic table.
  - b. method for determining atomic number.
  - c. test for radioactivity.
  - d. use of X rays.
- \_\_\_\_\_ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar
  - a. properties.
  - b. atomic numbers.
  - c. densities.
  - d. colors.
- \_\_\_\_\_ 7. In developing his periodic table, Mendeleev listed on cards each element's name, atomic mass, and
  - a. atomic number.
  - b. electron configuration.
  - c. isotopes.
  - d. properties.
- \_\_\_\_\_ 8. Mendeleev predicted that the spaces in his periodic table represented
  - a. isotopes.
  - b. radioactive elements.
  - c. permanent gaps.
  - d. undiscovered elements.
- \_\_\_\_\_ 9. Mendeleev's table was called periodic because the properties of the elements
  - a. showed no pattern.
  - b. occurred at repeated intervals called periods.
  - c. occurred at regular time intervals called periods.
  - d. were identical.
- \_\_\_\_\_ 10. The person whose work led to a periodic table based on increasing atomic number was
  - a. Moseley.
  - b. Mendeleev.
  - c. Rutherford.
  - d. Cannizzaro.
- \_\_\_\_\_ 11. Moseley's work led to the realization that elements with similar properties occurred at regular intervals when the elements were arranged in order of increasing
  - a. atomic mass.
  - b. density.
  - c. radioactivity.
  - d. atomic number.
- \_\_\_\_\_ 12. Who used his experimental evidence to determine the order of the elements according to atomic number?
  - a. Meyer
  - c. Stas

# Modern Chemistry Chapter 5 Review Answer Key

**J Ma**



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