

## 7-3 Review and Reinforcement

### Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

Number	Prefix
1	mono
2	di
3	tri
4	tetra
5	penta
6	hexa
7	hepta
8	octa
9	nona
10	deca

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

cation  
anion

hydrates  
anhydrates

electronegative  
hydrogen ions

ionic  
molecular

- Ionic compounds that absorb water into their solid structures form substances called **—hydrate—**.
- When naming an ionic substance, the name of the **—cation—** comes first.
- To name a binary molecular compound, add the suffix *-ide* to the more **—electronegative—** of the two elements.
- By one definition, an acid is a molecular compound that dissolves in water to produce **—hydrogen ions—**.
- The name of an acid is usually derived from the name of its characteristic **—anion—**.
- Prefixes are often used when naming **molecular** compounds.

# Modern Chemistry Chapter 7 Review Answers

**Ralph Thomas Myers**



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**American Chemical Review**, 1885 *Modern Chemistry* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

**Principles of Modern Chemistry** David W. Oxtoby, H. P. Gillis, Norman H. Nachtrieb, 2002 This successful textbook continues to address students in honours and high mainstream general chemistry courses The chapters in the fourth edition are organised within larger units to emphasise the conceptual structure of chemistry

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**General Chemistry** Ralph H. Petrucci, 1989 *Modern University Chemistry, Custom Pub* Norbert T. Porile, 1993

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