

Chapter Test A

Chapter: Atoms: The Building Blocks of Matter

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. Neutral atoms contain equal numbers of
- electrons and neutrons.
 - protons and neutrons.
 - protons and electrons.
 - protons, electrons, and neutrons.
- _____ 2. The atomic symbol for beryllium, ${}^9_4\text{Be}$, indicates that the
- atomic number is 4.
 - atomic number is 9.
 - mass number is 4.
 - atomic number is equal to $9 - 4$.
- _____ 3. The atomic mass of an atom of carbon is 12 amu, and the atomic mass of an atom of oxygen is 16 amu. To produce CO, 16 g of oxygen combine with 12 g of carbon. To produce CO_2 , 32 g of oxygen combine with 12 g of carbon. The ratio of the mass of oxygen in CO_2 to the mass of oxygen in CO is
- 1:1.
 - 2:1.
 - 1:2.
 - 8:3.
- _____ 4. Rutherford's gold-foil experiment led him to conclude that
- Thomson's plum pudding model of the atom was accurate.
 - alpha particles were a poor choice for a bombardment material.
 - a dense region of positive charge existed somewhere in the atom.
 - light was emitted by electrons returning to ground state.
- _____ 5. Which of the following is *not* one of the five principles of Dalton's theory?
- Atoms of different elements combine in simple, whole-number ratios to form compounds.
 - All matter is made of indivisible, indestructible atoms.
 - All atoms have similar physical and chemical properties.
 - Chemical reactions consist of the combination, separation, or rearrangement of atoms.

Modern Chemistry Chapter Atoms Test Answers

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