#### PRE TEST

#### Chapter 6 Test-General/Honors Chemistry

#### Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

C 2.2 \_\_\_\_\_\_1. cation a. a positively charged ion C2.2 8 2. electronegativity b. an element whose highest occupied s or p sublevels are partially filled C.2.5 3. inner transition metal c. the energy required to remove an electron from an atom C.2.6 4. alkali earth metal d. good conductor of heat and electric current. C.2.6 h 5. noble gas e. any element whose highest occupied s sublevel and a nearby d sublevel contain electrons C.2.5 6. representative element f. any nonmetal in group 7A c 2 2 \_\_\_\_\_ 7. metal g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) C.2.6 f 8. halogen h. an element in which the highest occupied s and p sublevels are completely filled C2.5 \_\_\_\_\_\_9. transition metal i. an element in group 2A C.2.2 c 10. ionization energy an element whose highest s sublevel and a nearby f sublevel contain electrons

#### Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- C.2.6 \_\_b \_\_11. The alkali metals do not include
  a. l.i c. Na
  b. Ca d. Rb
- C.2.6 \_\_\_\_\_12. The element iodine, I, is a a. period S alkali metal b. period 4 halogen

c. period 5 halogen d. period 5 transition metal

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**Minjie Lin** 

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