

## Chapter Test B

### Chapter: Chemical Equations and Reactions

Write a concise answer (possibly with the use of the text or phrase structure completed with appropriate nouns and verbs) to each question.

- \_\_\_\_\_ 1. The product(s) of a slightly exothermic combustion is a double displacement reaction. Write the balanced equation of it.
  - a. **no**
  - b. **propene**
  - c. **carbon monoxide**
  - d. **hydrogen**
- \_\_\_\_\_ 2. The balance chemical equation is experimentally to follow the
  - a. **law of conservation of mass**
  - b. **conservation**
  - c. **conservation of the particles**
  - d. **conservation of products**
- \_\_\_\_\_ 3. In chemical equation, the coefficient of  $\text{H}_2$  in the following equation is
  - a. **three**
  - b. **one**
  - c. **two**
  - d. **unbalanced**
- \_\_\_\_\_ 4. The following two displacement reactions are given here. In them
  - a. **silver metal reacts with chlorine gas to produce silver chloride**
  - b. **hydrogen gas reacts with oxygen gas to produce water**
  - c. **hydrogen gas reacts with oxygen gas to produce hydrogen peroxide**
  - d. **hydrogen gas reacts with oxygen gas to produce water and hydrogen peroxide**
- \_\_\_\_\_ 5. The coefficient of the chemical equation
  - a. **indicates the number of moles of each substance that react**
  - b. **shows the number of grams of each substance that react**
  - c. **shows the number of moles of the substance**
  - d. **shows the number of moles of each atom**
- \_\_\_\_\_ 6. If  $\text{H}_2$  is shown that means  $\text{H}_2$  is the gas variable, then
  - a. **it will replace some of the electrons**
  - b. **it will replace some of the electrons**
  - c. **it will replace some of the electrons**
  - d. **it will replace some of the electrons**

# Modern Chemistry Chapter Test B Answers

**TD Snyder**



## Modern Chemistry Chapter Test B Answers:

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