

Chemistry Chapter 5 Exam

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- _____ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to
 - a. Mendeleev.
 - b. Moseley.
 - c. Bohr.
 - d. Ramsay.
- _____ 2. Mendeleev left spaces in his periodic table and predicted several elements and their
 - a. atomic numbers.
 - b. colors.
 - c. properties.
 - d. radioactivity.
- _____ 3. Mendeleev attempted to organize the chemical elements based on their
 - a. symbols.
 - b. properties.
 - c. atomic numbers.
 - d. electron configurations.
- _____ 4. Mendeleev noticed that properties of elements usually repeated at regular intervals when the elements were arranged in order of increasing
 - a. atomic number.
 - b. density.
 - c. reactivity.
 - d. atomic mass.
- _____ 5. Mendeleev is credited with developing the first successful
 - a. periodic table.
 - b. method for determining atomic number.
 - c. test for radioactivity.
 - d. use of X rays.
- _____ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar
 - a. properties.
 - b. atomic numbers.
 - c. densities.
 - d. colors.
- _____ 7. In developing his periodic table, Mendeleev listed on cards each element's name, atomic mass, and
 - a. atomic number.
 - b. electron configuration.
 - c. isotopes.
 - d. properties.
- _____ 8. Mendeleev predicted that the spaces in his periodic table represented
 - a. isotopes.
 - b. radioactive elements.
 - c. permanent gaps.
 - d. undiscovered elements.
- _____ 9. Mendeleev's table was called periodic because the properties of the elements
 - a. showed no pattern.
 - b. occurred at repeated intervals called periods.
 - c. occurred at regular time intervals called periods.
 - d. were identical.
- _____ 10. The person whose work led to a periodic table based on increasing atomic number was
 - a. Moseley.
 - b. Mendeleev.
 - c. Rutherford.
 - d. Cannizzaro.
- _____ 11. Moseley's work led to the realization that elements with similar properties occurred at regular intervals when the elements were arranged in order of increasing
 - a. atomic mass.
 - b. density.
 - c. radioactivity.
 - d. atomic number.
- _____ 12. Who used his experimental evidence to determine the order of the elements according to atomic number?
 - a. Meyer
 - c. Stas

Modern Chemistry Chapter 5 Section 3 Review Answers

ML Morrison



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