

## CHAPTER 8 REVIEW

# Chemical Equations and Reactions

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**SHORT ANSWER** Answer the following questions in the space provided.

2. \_\_\_\_\_ According to the law of conservation of mass, the total mass of the reacting substances is

- (a) always more than the total mass of the products.
- (b) always less than the total mass of the products.
- (c) sometimes more and sometimes less than the total mass of the products.
- (d) always equal to the total mass of the products.

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

\_\_\_\_\_ 2. In the reaction described by the word equation



- a. sodium hydroxide is a product.
- b. sodium is a product.
- c. water is a reactant.
- d. Both (a) and (c).

\_\_\_\_\_ 3. Which of the following indicates that a chemical equation is balanced?

- a. The numbers of atoms of each element are the same on both sides of the equation.
- b. All of the coefficients are the same.
- c. The numbers of molecules on each side of the equation are equal.
- d. The sums of the coefficients on each side of the equation are equal.

\_\_\_\_\_ 4. From a complete and correctly written chemical equation, you can obtain the

- a. chemical formulas of the reactants and products.
- b. relative amounts of the reactants and products.
- c. physical states of the reactants and products.
- d. All of the above.

\_\_\_\_\_ 5. To balance a chemical equation, you must

- a. adjust the subscripts.
- b. adjust the number of products.
- c. adjust the number of reactants.
- d. adjust the coefficients.

\_\_\_\_\_ 6. The formation of a solid product in a chemical reaction is represented by the symbol

- a. (aq).
- b. (g).
- c. (l).
- d. (s).

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**EW Minium**



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*Medicinal Chemistry* Norma K Dunlap, Donna M Huryn, 2025-11-28 This book provides a concise introduction to the medicinal chemistry of therapeutically active compounds. Written in an accessible style it offers a brief overview of the drug development process from discovery to approved drug. Emphasizing medicinal chemistry strategies the content is organized by target as well as therapeutic area drug classes. It covers existing drugs and the discovery process that led to their development. Aimed at students and professionals interested in drug discovery. Medicinal Chemistry requires little prior knowledge making it ideal for classroom and training seminar use. Written by experts in the field this outstanding new edition explains medicinal chemistry on chemical mechanism terms allowing for a more interesting and user friendly text. Includes valuable insights toward the various pathways taken by the pharmaceutical industry in drug discoveries. Enhanced text includes questions raised and suggestions made from students in the authors medicinal chemistry class. Content provided to enhance the text includes a solutions manual, PowerPoint slides for lecture outlines and especially for the earlier chapters a selection of videos. An accompanying website will be updated to include additional material. 1 Journal Club exercises current journal articles with guiding questions for student analysis two per chapter. 2 additional case studies. 3 organic synthesis analysis selected syntheses both discovery and process will be discussed and 4 updates on current advances in the field of medicinal chemistry. The book benefits a diverse audience including upper level undergraduates and graduates who are studying in the fields of medicinal chemistry and drug discovery as well as scientists working in the pharmaceutical industry.

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