

Molality

The number of moles of a solute per mass of a solvent is referred to as molality

$$\text{Molality (M)} = \frac{\text{number of moles of solute}}{\text{mass of solvent in kg}}$$

$$\text{Molality (M)} = \frac{g \times 1000}{W \times m}$$

Molality M Chemistry If8766

Kathleen Armour



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