

# Molar Mass

# Worksheet-Answer Key

Name	Date

Exercise 1. How much does one mole weigh (molar mass/molecular weight) for the following? Show work: which numbers are added/multiplied, how many decimal places are used from the periodic table, and your result.

Na	23g	NaF	42g
	22.99 → 23 g		23 + 19 = 42  g
Zn	65.4 g _	СО	28 g
	65.38 → 65.4 g		12 + 16 = 28  g
solid iron	55.9 g _	$CO_2$	44 g
	Fe 55.85 → 55.9 g	12	$+2 \times 16 = 12 + 32 = 44 \text{ g}$
Ca <sup>2+</sup> (aq)	40 g	$CH_4$	16 g
	same as Ca 40.08 → 40 g		$12 + 4 \times 1 = 12 + 4 = 16$
CaCO <sub>3</sub>	100 g _	$C_4H_{10}$	58 g
	$40 + 12 + 3 \times 16 = 100g$		$4 \times 12 + 10 \times 1 = 58g$

# **Molar Mass Of Butane Worksheet Answers**

**Disha Experts** 

**Molar Mass Of Butane Worksheet Answers:** 

Unveiling the Magic of Words: A Overview of "Molar Mass Of Butane Worksheet Answers"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Molar Mass Of Butane Worksheet Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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