

The Molar Volume of a Gas

In this experiment, you will determine the molar volume of a gas by conducting a chemical reaction that produces a gas, as shown in the reaction equation below.



You will react a known mass of solid magnesium with an excess of hydrochloric acid, in a sealed vessel, and use the pressure change to calculate molar volume at STP.

OBJECTIVES

In this experiment, you will

- Measure the gas production of a chemical reaction by a pressure change.
- Determine the molar volume of the gas produced in the reaction.
- Calculate the molar volume of a gas at STP.



Figure 1

MATERIALS

Vernier computer interface
computer
Vernier Gas Pressure Sensor
Temperature Probe
600 mL or one liter beaker
10 mL graduated cylinder

magnesium ribbon, pre-cut
3.0 M HCl solution
small beaker for HCl solution
ruler
125 mL Erlenmeyer flask
syringe

Molar Volume Of Hydrogen Gas Lab Answers

RS Peters



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