

Chemistry: Molarity



Directions: Using these formulas, solve the following problems:

molarity = mol solute / L solution

1 amu = 1 gram / mol

$M_1V_1 = M_2V_2$

Problem 1: Calculate the molarity of a 2.21 L solution made with 31.7 g of calcium chloride. Answer: _____

Problem 2: How many grams of NaCl would be required to produce a 6.1 M solution with a volume of 0.64 L? Answer: _____

Problem 3: How many grams of NaOH would be required to produce a 5.36 M solution with a volume of 1.37 L? Answer: _____

Problem 4: How many grams of KCl would you need to weigh out to prepare 0.416 L of a 3.37 M KCl solution? Answer: _____

Problem 5: Calculate the molarity of a 6.68 mL solution of 6.89 g NaCl. Answer: _____

Problem 6: Calculate the volume in mL if you have 10.9 g of Na_2CO_3 from a 0.112 M solution. Answer: _____

Problem 7: Calculate the volume in mL if you have 28.1 g LiOH from a 7.11 M solution. Answer: _____

Problem 8: Calculate the final concentration if water is added to 0.789 L of a 3.68 M HCl solution to give a volume of 0.387 L. Answer: _____

Problem 9: Calculate the final concentration if a 8.63 mL of a 1.12 M HCl solution is diluted with water to 2.54 mL. Answer: _____

Problem 10: Calculate the final concentration if water is added to 0.235 L of a 6.11 M KBr solution to give a volume of 5.79 L. Answer: _____

Molarity Worksheet Answers Instructional Fair

RS Peters



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Whispering the Secrets of Language: An Emotional Journey through **Molarity Worksheet Answers Instructional Fair**

In a digitally-driven earth where monitors reign great and immediate communication drowns out the subtleties of language, the profound techniques and mental subtleties concealed within phrases often move unheard. However, located within the pages of **Molarity Worksheet Answers Instructional Fair** a charming fictional value blinking with organic feelings, lies an exceptional journey waiting to be undertaken. Penned by an experienced wordsmith, this marvelous opus invites viewers on an introspective journey, delicately unraveling the veiled truths and profound impact resonating within the cloth of every word. Within the emotional depths of this moving evaluation, we can embark upon a sincere exploration of the book is key subjects, dissect its fascinating publishing model, and fail to the strong resonance it evokes heavy within the recesses of readers hearts.

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