

# MOLARITY BY DILUTION

79 + 52 = 131 Name \_\_\_\_\_

Acids are usually acquired from chemical supply houses in concentrated form. These acids are diluted to the desired concentration by adding water. Since moles of acid before dilution = moles of acid after dilution, and moles of acid =  $M \times V$  then,  
 $M_1 \times V_1 = M_2 \times V_2$  Solve the following problems.

1. How much concentrated 18 M sulfuric acid is needed to prepare 250 mL of a 6.0 M solution?

$$V_1 = \frac{M_2 V_2}{M_1} = \frac{(6.0 M)(250 \text{ mL})}{18 M}$$

83 mL

2. How much concentrated 12 M hydrochloric acid is needed to prepare 100 mL of a 2.0 M solution?

$$V_1 = \frac{M_2 V_2}{M_1} = \frac{(2.0 M)(100 \text{ mL})}{12 M}$$

17 mL

3. To what volume should 25 mL of 15 M nitric acid be diluted to prepare a 3.0 M solution?

$$V_2 = \frac{M_1 V_1}{M_2} = \frac{(15 M)(25 \text{ mL})}{3.0 M}$$

125 mL

4. To how much water should 50. mL of 12 M hydrochloric acid be added to produce a 4.0 M solution?

$$V_2 = \frac{M_1 V_1}{M_2} = \frac{(12 M)(50 \text{ mL})}{4.0 M} = 150 \text{ mL} - 50 \text{ mL}$$

100 mL H<sub>2</sub>O

5. To how much water should 100. mL of 18 M sulfuric acid be added to prepare a 1.5 M solution?

$$V_2 = \frac{M_1 V_1}{M_2} = \frac{(18 M)(100 \text{ mL})}{1.5 M} = 1200 \text{ mL} - 100 \text{ mL} = 1100 \text{ mL}$$

1.1 L

# Molarity Instructional Fair

**Wenbin Ji**



## **Molarity Instructional Fair:**

*Science Up to Standards, Grades 5 - 8* Pam Walker, Elaine Wood, 1999-01-15 Science Up to Standards is loaded with activities based on the guidelines defined by the National Science Education Standards The format follows closely those standards which encourage teachers to provide students with activities requiring both partial and full inquiries In partial inquiry activities students are given some instructions on how to approach the problem solving issue In full inquiry activities students design their own strategies for finding answers This great resource provides activities in seven content areas Science as Inquiry Physical Science Life Science Earth and Space Science Science and Technology Science in Personal and Social Perspectives and History and Nature of Science *Computers in Chemical Education and Research* E.

Ludena, 2013-03-09 The impact of computers on all realms of Chemistry has been one of the most important factors in the development of this science during the last years In recognition of this fact in 1971 the First International Conference on Computers in Chemical Research and Education was held at DeKalb Illinois USA A second Conference took place in Ljubljana Yugoslavia in 1973 and this third Conference in Caracas Venezuela in 1976 The aim of these conferences was to provide a high level forum for the leading researchers to exchange information at the frontiers of present day computer applications to the different fields of Chemistry The present Third International Conference on Computers in Chemical Research Education and Technology whose proceedings are published in the present volume was conceived by means of a series of invited lectures as a survey of the present state of the art in some of the most relevant areas of computer applications in Chemistry *The Education Index*, 1985 **Current Index to Journals in Education**, 1985

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