

Mole Ratios

How can the coefficients in a chemical equation be interpreted?

Why?

A balanced chemical equation can tell us the number of reactant and product particles (ions, atoms, molecules or formula units) that are necessary to conserve mass during a chemical reaction. Typically when we balance the chemical equation we think in terms of individual particles. However, in real life the reaction represented by an equation occurs an unimaginable number of times. Short of writing very large numbers (10^{23} or larger) in front of each chemical in the equation, how can we interpret chemical equations so that they more realistically represent what is happening in real life? In this activity you will explore the different ways a chemical reaction can be interpreted.

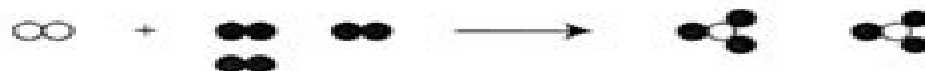
Model 1 – A Chemical Reaction



1. Consider the reaction in Model 1.
 - a. What are the coefficients for each of the following substances in the reaction?



- b. Draw particle models below to illustrate the reaction in Model 1.



2. Consider each situation below as it relates to the reaction in Model 1.
 - a. Calculate the amount of reactants consumed and products made.
 - b. Record the ratio of N_2 to H_2 to NH_3 . Reduce the ratio to the lowest whole numbers possible.

	N_2 Consumed	H_2 Consumed	NH_3 Produced	Ratio $\text{N}_2:\text{H}_2:\text{NH}_3$ (reduced)
For a single reaction, how many molecules of each substance would be consumed or produced?	1 molecule	3 molecules	2 molecules	1:3:2
If the reaction occurred one hundred times, how many molecules would be consumed or produced?	100 molecules	300 molecules	200 molecules	1:3:2
If the reaction occurred 538 times, how many molecules would be consumed or produced?	538 molecules	1614 molecules	1076 molecules	1:3:2

Mole Ratio Pogil Key Why

Thomas Bond, Chris Hughes



Mole Ratio Pogil Key Why:

An Introduction to Chemistry Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life. It provides a major overview of topics typically found in first year chemistry courses in the USA. The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box. Numerous examples are presented in every chapter to aid students and provide helpful self learning tools. The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry. *The Practice of Chemistry Study Guide & Solutions Manual* Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes. Also includes solutions and explanations to odd numbered exercises. **The IIT Foundation Series - Chemistry Class 9, 2/e**, **Teaching Science for**

Understanding James Joseph Gallagher, 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities. **Nature Of Chemistry Volume - 2** Aditya Tripathi, **A-level Chemistry Challenging Drill Questions (Concise) (Yellowreef)** Thomas Bond, Chris Hughes, 2013-11-04 questions from top schools colleges since 2003 complete encyclopedia of over 400 questions complete answer keys full set of step by step solution approaches available separately Complete edition and concise edition eBooks available [E3 Chemistry AP Exam Practice - 2018: With Answers, Explanations and Scoring Guidelines](#) Effiong Eyo, 2018-01-15 Preparing for Chemistry AP Exam has never been easier more enticing more exciting more engaging more understandable and less overwhelming. Our book is written to help students do more know more and build confidence for a higher mark on their AP exam. With a total of four practice tests with answers and explanations this book can be used as a primary question practice resource or as a supplementary resource to other AP chemistry book. Book Summary Organized engaging doable quick practice quality question sets Clear brief simple and easy to understand correct answer explanations With scoring guidelines to all free response questions Start your Chemistry AP Exam Practice today Good Luck AP is a trademark registered by the College Board which is not affiliated with and does not endorse this book. [Proceedings of the ... National Conference on Hazardous Wastes and Hazardous Materials](#), 1987

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