

Mole Ratio Worksheet

For each equation provided, determine the following molar ratios.



$\text{N}_2 : \text{H}_2 =$

$\text{NH}_3 : \text{H}_2 =$



$\text{CH}_4 : \text{O}_2 =$

$\text{O}_2 : \text{H}_2\text{O} =$



$\text{O}_2 : \text{SO}_3 =$

$\text{O}_2 : \text{SO}_2 =$



$\text{Fe} : \text{Fe}_2\text{O}_3 =$

$\text{O}_2 : \text{Fe}_2\text{O}_3 =$



$\text{PCl}_3 : \text{Cl}_2 =$

$\text{PCl}_3 : \text{PCl}_5 =$



$\text{Mg} : \text{Mg}_3\text{P}_2 =$

$\text{P}_4 : \text{Mg}_3\text{P}_2 =$



$\text{NH}_3 : \text{N}_2 =$

$\text{H}_2\text{O} : \text{O}_2 =$



$\text{MgCl}_2 : \text{Al} =$

$\text{Al} : \text{AlCl}_3 =$



$\text{CO} : \text{CO}_2 =$

$\text{Fe} : \text{CO} =$



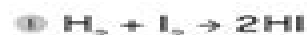
$\text{WO}_3 : \text{W} =$

$\text{H}_2 : \text{H}_2\text{O} =$



$\text{Al} : \text{AlCl}_3 =$

$\text{HCl} : \text{H}_2 =$



$\text{H}_2 : \text{HI} =$

$\text{I}_2 : \text{HI} =$

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Patrick Vollmar

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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