

# STOICHIOMETRY: MASS – MOLE WORKSHEET

Solve the following stoichiometry problems and write the correct answer with units (For example: 34 mol ó 4.62 g) Round up to 2 decimals.

1. Based on the following chemical equation, please answer the following questions:



- a. Calculate the amount of moles of NaCl needed to produce 75g of FeCl<sub>3</sub>.

Answer:

- b. Calculate the amount of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> in grams to produce 215 g of Na<sub>2</sub>SO<sub>4</sub>.

Answer:

- c. Calculate the amount in grams of FeCl<sub>3</sub> produced with 95g of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>

Answer:

- d. Calculate the amount of moles of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> needed to react with 123.5 g of NaCl

Answer:

2. How many moles of CaCO<sub>3</sub> can be produced if we make 12.3g of Ca(HCO<sub>3</sub>)<sub>2</sub> reacts with Ca(OH)<sub>2</sub>?



Answer:

# Mole To Mass Stoichiometry Problems Answers

**Charles H. Corwin**



## **Mole To Mass Stoichiometry Problems Answers:**

*The Practice of Chemistry Study Guide & Solutions Manual* Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

*Chemistry* Neil D. Jespersen, Alison Hyslop, 2021-12-03 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

**Basics for Chemistry** David A. Ucko, 2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material

Foundations of College Chemistry Morris Hein, Susan Arena, 2010-01-26 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis

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chemistry and biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career

**The Practice of Chemistry** Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink [http://www.bfwpub.com/pdfs/wink/POCPowerPoint\\_Final.ppt](http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt) 832KB

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CHEMISTRY SILBERBERG, 2003

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Introductory Chemistry Kevin Revell, 2020-11-17 Introductory Chemistry creates light bulb moments for students and provides unrivaled support for instructors Highly visual interactive multimedia tools are an extension of Kevin Revell's distinct author voice and help students develop critical problem solving skills and master foundational chemistry concepts necessary for success in chemistry

**Improving Student Comprehension in Chemistry Laboratories** Tracy Lynn Haroff, 2006

Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook McGraw Hill, 2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change

**General Chemistry, Reactions First** Kevin Revell, 2024-12-04 Revell's General Chemistry empowers students to grasp essential topics and concepts with more ease Using a friendly approach the text uses metaphors and relatable examples to demystify even the most challenging subjects in general chemistry

Prentice Hall Chemistry, 2000

Chemistry James E. Brady, John R. Holum, 1993-01-21 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping students develop skills in analytical thinking and problem solving Students are taught in a variety of ways to think of skills as tools that can be used to solve complex problems Several aids are included to help focus and inspire student interest frequent reference to common chemicals in commercial products numerous photographs of reactions in chapter practice exercises following worked examples

The Software Encyclopedia 2000 Bowker Editorial Staff, 2000-05

*Merrill Chemistry* Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

*Student Solutions Manual: Ssm Chemistry* Deborah Wiegand, 2003 This

manual contains complete worked out solutions to all follow up problems and about half of all the chapter problems Each chapter of solutions opens with a summary of the text chapter content and a list of key equations needed to solve the problems

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