Phet - Molecular Polarity

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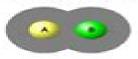
Part 1: TWO ATOMS

Select the "Two Atoms" tab at the top of the page and then make sure the following check boxes are turned on:

☑ Bond dipole, ☑ Partial Charges, and ☑ Electron Density.

In the "Electronegativity" boxes at the top, make Atom A "less" and Atom B "less".





The gray shading shows the distribution of electrons. When do you notice about the distribution of electrons when **both** atoms have LOW Electronegativity?

- In the "Electronegativity" boxes at the top, keep Atom A "less" and gradually make Atom B "more".
- a. What do you notice about the electron distribution as you make Atom B more electronegative?
- b. Where are the partial charges located? Explain why.
- c. Which way is the bond vector pointing?
- 3. In the "Electronegativity" boxes at the top, make Atom A "less" and gradually keep Atom B "more".
- Describe the electron distribution and the partial charges when <u>both</u> atoms have high electronegativity.
- b. Explain why you think this happened.

Part 2: THREE ATOMS

Select the "Two Atoms" tab at the top of the page and then make sure the following check boxes are turned on:

☑ Bond dipole, ☑ Molecular Dipole, and ☑ Partial Charges.

Draw the partial charges, bond dipoles, and molecular dipoles for each situation.



Molecule Polarity Phet Lab

Peter J. W. Debye

Molecule Polarity Phet Lab:

Simulations and Student Learning Matthew Schnurr, Anna MacLeod, 2021-01-04 The book underlines the value of simulation based education as an approach that fosters authentic engagement and deep learning Performance Assessment: Concepts, Methodologies, Tools, and Applications Management Association, Information Resources, 2019-10-11 As teaching strategies continue to change and evolve and technology use in classrooms continues to increase it is imperative that their impact on student learning is monitored and assessed New practices are being developed to enhance students participation especially in their own assessment be it through peer review reflective assessment the introduction of new technologies or other novel solutions Educators must remain up to date on the latest methods of evaluation and performance measurement techniques to ensure that their students excel Learning and Performance Assessment Concepts Methodologies Tools and Applications is a vital reference source that examines emerging perspectives on the theoretical and practical aspects of learning and performance based assessment techniques and applications within educational settings Highlighting a range of topics such as learning outcomes assessment design and peer assessment this multi volume book is ideally designed for educators administrative officials principals deans instructional designers school boards academicians researchers and education students seeking coverage on an educator s role in evaluation design and Handbook of Research on Innovative Pedagogies and Technologies for analyses of evaluation methods and outcomes Online Learning in Higher Education Vu, Phu, Fredrickson, Scott, Moore, Carl, 2016-12-28 The integration of technology has become an integral part of the educational environment By developing new methods of online learning students can be further aided in reaching goals and effectively solving problems The Handbook of Research on Innovative Pedagogies and Technologies for Online Learning in Higher Education is an authoritative reference source for the latest scholarly research on the implementation of instructional strategies tools and innovations in online learning environments Featuring extensive coverage across a range of relevant perspectives and topics such as social constructivism collaborative learning and projects and virtual worlds this publication is ideally designed for academicians practitioners and researchers seeking current research on best methods to effectively incorporate technology into the learning environment The Big Book of **Chemistry Teacher Stories** Jeff Lark, Stories from years of teaching high school chemistry Amino-acid, Peptide & Protein Abstracts ,1979 **Polar molecules** Peter J. W. Debye, 1970

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