

# Molecular Polarity

Molecules will be polar if

a) bonds are polar

AND

b) the molecule is NOT "symmetric"

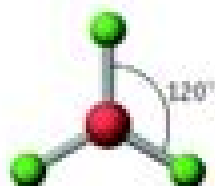
Linear



$AX_2$

Example:  $BeF_2$

Trigonal-planar



$AX_3$

Example:  $BF_3$

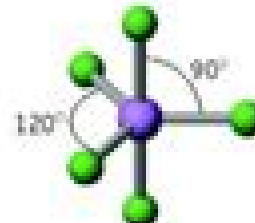
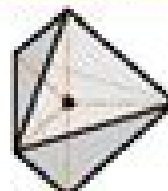
Tetrahedral



$AX_4$

Example:  $CF_4$

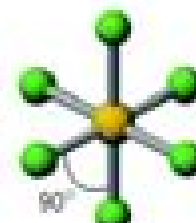
Trigonal-bipyramidal



$AX_5$

Example:  $PF_5$

Octahedral



$AX_6$

Example:  $SF_6$

All above are **NOT** polar

# Molecule Polarity Pogil

**P. Debye**



## **Molecule Polarity Pogil:**

**Analytical Chemistry** Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the Beer-Lambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more

**Polar Molecules** Peter Josef William Debye, 1958 [Polar Molecules](#) P. Debye, 1965

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