

NOMENCLATURE PACKET

Worksheet I: Binary Ionic Compounds (representative metals)

1. Name the following ionic compounds:

- a. Al_2O_3 Aluminum oxide
- b. Cs_2O Cesium oxide
- c. Rb_3N Rubidium nitride
- d. Ca_3N_2 Calcium nitride
- e. $SrSe$ Strontium selenide
- f. Cs_2S Cesium sulfide
- g. Al_2S_3 Aluminum sulfide
- h. $LiBr$ Lithium bromide
- i. Mg_3N_2 Magnesium nitride
- j. CaF_2 Calcium fluoride

2. Write the chemical formula for the following ionic compounds:

- a. Barium nitride Ba_3N_2
- b. Indium Fluoride InF_3
- c. Calcium oxide CaO
- d. Sodium nitride Na_3N
- e. Magnesium chloride $MgCl_2$
- f. Potassium oxide K_2O
- g. Magnesium oxide MgO
- h. Potassium sulfide K_2S
- i. Lithium nitride Li_3N
- j. Strontium fluoride SrF_2
- k. Aluminum sulfide Al_2S_3
- l. Calcium oxide CaO

Ionic Nomenclature Review Vocabulary Answer Key

**Philip C. Keller, Raymond Chang, Jill L.
Keller**

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