

Lesson 24: Ionic Bonds

Content Standard

The learners demonstrate an understanding of ionic bond formation in terms of atomic properties.

Performance Standards

The learners shall be able to identify ionic compounds used in daily life.

Learning Competencies

At the end of the lesson, the learners:

1. Relate the stability of noble gases to their electron configuration (STEM_GC11CB-11d-g-67);
2. State the Octet Rule (STEM_GC11CB-11d-g-68);
3. Determine the charge of ions formed by the representative elements and relate this to their ionization energy or electron affinity, valence electron configuration, and position in the periodic table. (STEM_GC11CB-11d-g-69);
4. Draw the Lewis structure of ions (STEM_GC11CB-11d-g-70);
5. Predict the formula of the ionic compound formed by a metal and non-metal among the representative elements. (STEM_GC11CB-11d-g-71);
6. Draw the Lewis structure of ionic compounds. (STEM_GC11CB-11d-g-72);
7. List the properties of ionic compounds and explain these properties in terms of their structure (STEM_GC11CB-11d-g-73); and
8. Perform exercises on writing Lewis structures of ions/ionic compounds and molecules. (STEM_GC11CB-11d-g-74).

Lesson Outline

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Enrichment	Ions in the Human Body	5
Evaluation	Worksheet	20

Resources

- (1) Chang, R. & Goldsby, K. (2016). Chemistry: (12th ed.). New York: McGraw-Hill.
- (2) Jostan, M., Castellon, M. & Hogg, J. (2007). The World of Chemistry: Essentials Belmont, CA: Brooks/Cole, Cengage Learning
- (3) Petrucci, Herring, Madura, and Blaconette (2011). General Chemistry and Modern Applications, 10th Ed. Pearson Canada, Inc.
- (4) Zumdahl, S.S. and Zumdahl, S.A (2013). Chemistry, 8th ed. Cengage Learning

Specific Learning Outcomes

At the end of the lesson, the learners shall be able to:

1. State the Octet Rule.
2. Draw the Lewis dot symbols of atoms and ions.
3. Identify elements that easily form cations and anions based on ionization energies and electron affinities
4. Explain the formation of ionic bonds using electron configuration and the Lewis dot symbols.
5. Relate lattice energy with physical properties of ionic compounds such as melting point.

Lesson 20 Covalenr Bonds Answer

Michael Brown

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