# ASSESSMENT

Section The Contrasponds to the indicated Contras Contrastes Con-

found of the Deginning of the Lebestron Exertise.

8. Cells that line cavities of the brain and secrete occubrospinal fluid

# Nervous Tissue and Nerves

March the terms in column A with the desi	criptions in column B. Place the letter of your choice in the space provided.
Column A	
a. Astrocyte -	Sheath of Schwann cell containing cytoplasm and nucleus that makes myelin
b. Avog	
c. Colliseral	2. Corresponds to rough endoplasmic reticulars to other cells
d. Dendrite	3. Network of threadnike structures within cell body and exercing into our
e. Myelin	4. Substance of Schwann cell composed of lipoproses that involves name
g. Neurofibrils	and increases impulse speed
h. Nissi bodies	5. Neuron process with many branches that conducts as action potential
(chromatophilic	(impulse) toward the cell body
aubstance) L. Unipolar neuron	6. Branch of an axon
	7. Star-shaped neuroglia between neurons and blood sessels
	Nerve fiber arising from a slight elevation of the cell body that conducts an action potential (impulse) away from the cell body
	Possesses a single process from the cell body
ART B: Assessments	
	criptions in column B. Place the letter of your choice in the space provides.
Column A	Column B
Effector	1. Transmits impulse from sensory to motor ocuron within cosmic across-
Ependymal cell	
Ganglion	2. Transmits impulse out of the brain or spiral cord to effection (woodes at
Interneuron (association neuron)	glands)
Microglia	granus)
Mosor (efferent) neuron	3. Transmits impulse into brain or spinal cord from receptors
ANYOR (CITCLEUR) DESILOG	Myelin-forming neuroglia in brain and spiral cord
This provides a second control of the second	
	5. Phagocytic neuroglia
Oligodendrocyte ensory (afferent) neuron	Separate capable of responding to motor impulse
	6. Structure capable of responding to motor the brein or spiral conf
	Phagocytic neuroglia     Structure capable of responding to motor impulse     Specialized mass of neuron cell bodies outside the brain or spiral cont     Specialized mass of the brain and secrete combrespinal fluid

# **Marieb Lab Manual Answer Key Exercise 6**

**Elaine Nicpon Marieb** 

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neutrons differ ions are atoms which contain an overall charge where number of protons number of electrons

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web aug 19 2022 atoms with the same number of protons but different numbers of neutrons are called isotopes they share almost the same chemical properties but differ in mass and therefore in physical properties there are stable isotopes which do not emit radiation and there are unstable isotopes which do emit radiation

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web atoms of the same element that differ in their numbers of neutrons are called isotopes many isotopes occur naturally usually one or two isotopes of an element are the most stable and common different isotopes of an element generally have the same physical and chemical properties because they have the same numbers of protons and electrons

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web isotopes are distinct nuclear species or nuclides as technical term of the same chemical element they have the same atomic number number of protons in their nuclei and position in the periodic table and hence belong to the same chemical element but differ in nucleon numbers mass numbers due to different numbers of neutrons in

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

isotopes worksheet answer key part i answer the questions - Feb 03 2022

web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

# how do isotopes differ in atomic structure example socratic - Sep 10 2022

web nov 27 2016 explanation basically isotopes are nuclei that have the same number of protons and are thus the same element but have unequal numbers of neutrons answer link the nucleus differs as a result of the number of neutrons in the atom the actual structure doesn t differ all that much but its mass does differ thus its density is a little solved isotopes differ from each other in what ways a they - Mar 04 2022

web chemistry questions and answers isotopes differ from each other in what ways a they have different numbers of electrons outside the nucleus b they have different atomic number o c they have different numbers of neutrons in the nucleus d more than one response is correct e they have different numbers of protons in the nucleus