

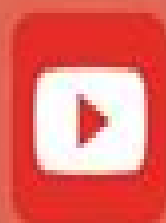
# IGCSE A\* MODEL ANSWERS

## 0620 CHEMISTRY

### MAY/JUNE - 2024

### PAPER - 61

### 0620\_S24\_QP\_61 (Q3-4)



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Table 1 shows the tests and the student's observations for solid P.

Tests	Observations
Test 1 Heat about half of solid P in boiling tube with test tube in fumes of $\text{HCl}$ .	<u>Amorphous white</u> The white solid forms a colourless liquid, steam effervesces from the boiling tube and condensation is seen at the top of the boiling tube. After a while the colourless liquid becomes a white solid.
Test 2 The remaining solid P is dissolved in water to form solution P. Solution P is divided into three portions. To the first portion of solution P, a glowing splint, with a glowing <u>blue</u> flame, is held over the mouth of the boiling tube and then is removed. Repeat the procedure and then bring red litmus paper at the mouth of the boiling tube.	<u>no <math>\text{H}_2</math> gas</u> <u>no <math>\text{H}_2\text{S}</math> gas</u> <u>no <math>\text{H}_2\text{SO}_4</math></u> <u>no change</u> The glowing red litmus paper remains red.
Test 3 To the second portion of solution P, add 1 cm <sup>3</sup> of dilute <u>nitric acid</u> followed by a few drops of <u>barium chloride solution</u> .	<u>white precipitate</u> $\text{SO}_4^{2-}$
Test 4 To the third portion of solution P, add a few drops of <u>acidified aqueous potassium manganate(VII)</u> .	The solution becomes light purple. $\text{SO}_3^{2-}$

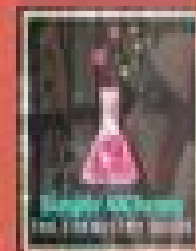
30. The observations in Test 1 show that solid P is hydrated.

31. Suggest a chemical test to show that the condensation at the top of the boiling tube contains water.

— anhydrous blue cobalt chloride paper

— pink  $\text{H}_2\text{O}$

— anhydrous white crumbly solid ✓



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**Xiang Xie**



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