

## CHEMISTRY A LEVEL MARK SCHEME

### ALDEHYDES AND KETONES

1.	<p>- <b>D</b> – It reacts with KCN in a <b>nucleophilic substitution reaction</b>. (1)</p> <p><b>EXPLANATION:</b></p> <p>A is clearly wrong. Benzene rings undergo electrophilic substitution and this is a cyclohexadiene ring. It could undergo <b>electrophilic addition</b> with HBr, so really, the fault here is with the word “substitution.”</p> <p>B is also wrong. <b>NaBH<sub>4</sub></b>, also referred to as tetrahydridoborate (III), is used in <b>reduction</b> reactions – if you were trying to make this molecule into an alcohol, you’d be all set! Nucleophilic addition can also occur using aldehydes, but commonly uses <b>HCN</b>. Nucleophilic addition-elimination (what the question mentions) however, is a completely different ballpark – it occurs when <b>nucleophiles react with acyl chlorides and acid anhydrides</b>.</p> <p>C is also wrong. <b>Ethanollic KOH</b> is used for <b>elimination</b>, and this molecule does have a C-Br bond. However, there is <b>no carbon-hydrogen bond adjacent</b> to the C-Br bond – both of the adjacent carbons have used up all of their bonds!</p> <p>D is the right answer. Because of the <b>C-Br bond</b>, a <b>negative cyanide ion</b> can attack the C-Br bond and <b>replace the bromine ion</b>.</p>
2.	<p>- <b>A</b> – <b>both can be formed by the alkaline hydrolysis of esters</b>. (1)</p> <p><b>EXPLANATION:</b></p>

# Mark Scheme Chemistry November 2paper 6

**Peipei Pang**



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