

Mark scheme

Chemistry Past Paper Questions 1

SECTION B: CHEMISTRY

Item	Expected answers	Additional guidance	Marks
4 (a)	5		1
4 (b)	An electron has a much smaller mass than a proton.		1
4 (c)	$(20 \times 10) + (80 \times 11)/100$ (1) = 10.8 (1)	$(200 + 80)/100$ (1) or $1080/100$ (1)	2
4 (d)	2 or 8 but not 3 for 1 mark. 2, 8, 3 for 2 marks.		2
Total mark			6

Item	Expected answers	Additional guidance	Marks
5 (a)	Any one of the following: It is on the right hand side of the periodic table (1) It is an inert gas/rare gas/noble gas (1)		1
5 (b)	Any two of the following: They have similar reactions to each other (1) They have similar properties to each other (1) They all have 1 electron in their outer shell (1)	Do not accept 'they are all alkali metals'	2
Total mark			3

Mark Scheme For 1m J 14

David Kirk



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