

**JIPMER-02-06-2019-Morning**  
**Chemistry**

1.  $^{19}\text{F}^{-1}$ ,  $^{16}\text{O}^{-2}$ ,  $^{20}\text{Ne}$  choose the correct statement
  - (1) Both  $\text{O}^{-2}$  and  $\text{F}^{-}$  are isoelectronic.
  - (2) All given have equal no of  $e^{-}$
  - (3)  $\text{F}^{-}$  and  $\text{Ne}$  have equal number of  $e^{-}$
  - (4) All
  
2. Hund's Rule state that:
  - (1) Number of two  $e^{-}$  can be in two separate orbitals
  - (2) Number of two  $e^{-}$  can be present with similar spin in a orbital.
  - (3) No one  $e^{-}$  can exist in 'd' orbital
  - (4) None of these
  
3. If two atoms have equal number of electron it is called:
  - (1) Iso-electronic
  - (2) isotone
  - (3) isobar
  - (4) None of these

# Jipmer 2mbbs Entrance Question Paper

**Padhraic Smyth**



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