

Mark scheme

Chemistry Past Paper Questions 1

SECTION B: CHEMISTRY

Item	Expected answers	Additional guidance	Marks
4 (a)	5		1
4 (b)	An electron has a much smaller mass than a proton.		1
4 (c)	$(20 \times 10) + (80 \times 11)/100$ (1) = 10.8 (1)	$(200 + 80)/100$ (1) or $1080/100$ (1)	2
4 (d)	2 or 8 but not 3 for 1 mark. 2, 8, 3 for 2 marks.		2
Total mark			6

Item	Expected answers	Additional guidance	Marks
5 (a)	Any one of the following: It is on the right hand side of the periodic table (1) It is an inert gas/rare gas/noble gas (1)		1
5 (b)	Any two of the following: They have similar reactions to each other (1) They have similar properties to each other (1) They all have 1 electron in their outer shell (1)	Do not accept 'they are all alkali metals'	2
Total mark			3

Mark Scheme Chemistry June 2013 Paper

Christian G. Meyer

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Mark Scheme Chemistry June 2013 Paper

1. Understanding the eBook Mark Scheme Chemistry June 2013 Paper
 - The Rise of Digital Reading Mark Scheme Chemistry June 2013 Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Mark Scheme Chemistry June 2013 Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mark Scheme Chemistry June 2013 Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mark Scheme Chemistry June 2013 Paper
 - Personalized Recommendations
 - Mark Scheme Chemistry June 2013 Paper User Reviews and Ratings
 - Mark Scheme Chemistry June 2013 Paper and Bestseller Lists

5. Accessing Mark Scheme Chemistry June 2013 Paper Free and Paid eBooks
 - Mark Scheme Chemistry June 2013 Paper Public Domain eBooks
 - Mark Scheme Chemistry June 2013 Paper eBook Subscription Services
 - Mark Scheme Chemistry June 2013 Paper Budget-Friendly Options
6. Navigating Mark Scheme Chemistry June 2013 Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Mark Scheme Chemistry June 2013 Paper Compatibility with Devices
 - Mark Scheme Chemistry June 2013 Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mark Scheme Chemistry June 2013 Paper
 - Highlighting and Note-Taking Mark Scheme Chemistry June 2013 Paper
 - Interactive Elements Mark Scheme Chemistry June 2013 Paper
8. Staying Engaged with Mark Scheme Chemistry June 2013 Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mark Scheme Chemistry June 2013 Paper
9. Balancing eBooks and Physical Books Mark Scheme Chemistry June 2013 Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mark Scheme Chemistry June 2013 Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mark Scheme Chemistry June 2013 Paper
 - Setting Reading Goals Mark Scheme Chemistry June 2013 Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mark Scheme Chemistry June 2013 Paper
 - Fact-Checking eBook Content of Mark Scheme Chemistry June 2013 Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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